

# Evaluations of Hourly PM Mass and Species Simulations by Models-3/CMAQ

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# Introduction

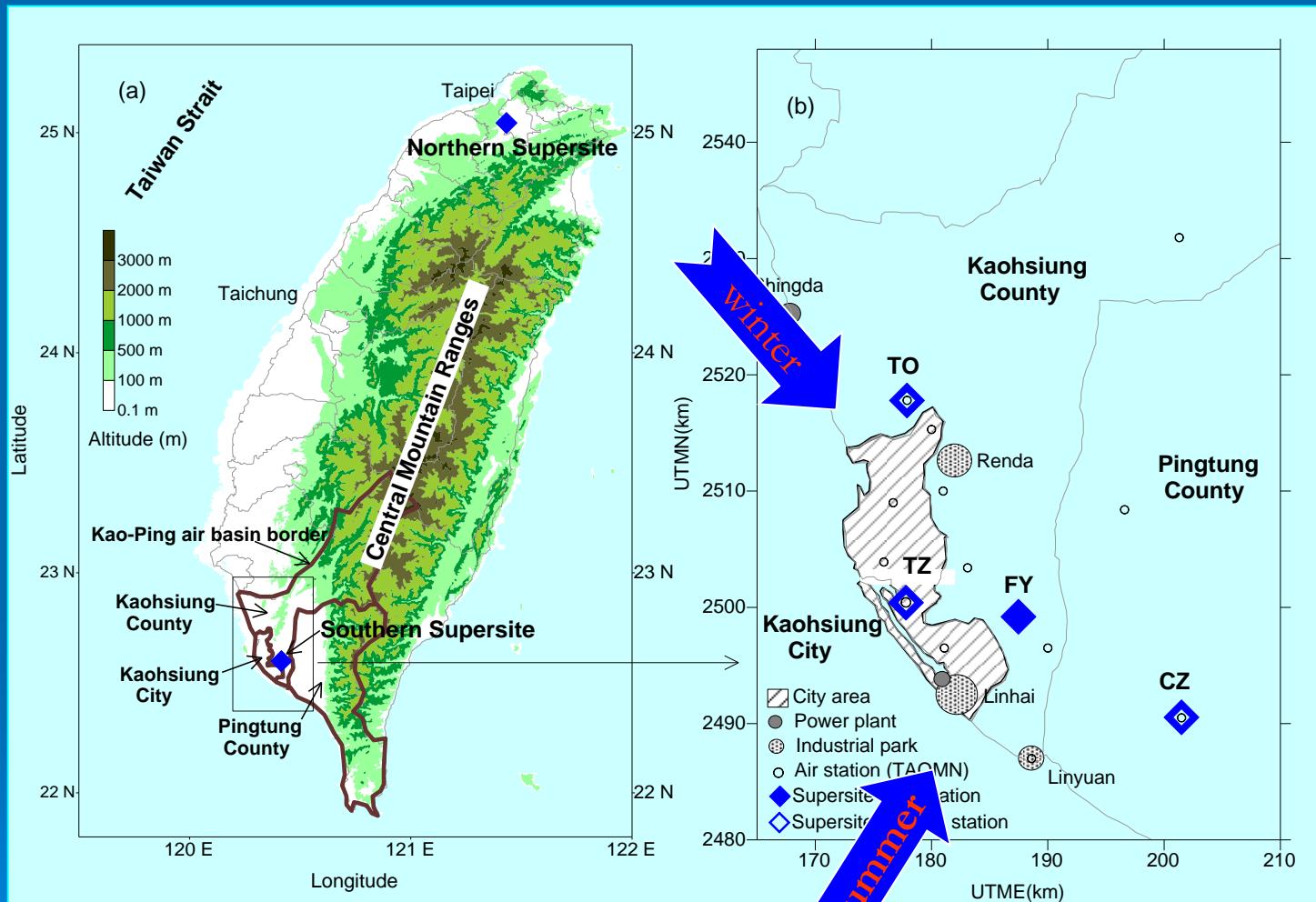
Model evaluation for PM simulation is based on 24-hour average concentrations for PM mass and species concentrations;

Model outputs and measurements are generally hourly concentrations.

	配對值絕對偏差	配對值常化偏差
PM <sub>10</sub>	< 150%	-50% ~ 80%
NOx	< 80%	-40% ~ 50%
SO <sub>2</sub>	< 80%	-40% ~ 50%

**Model evaluations based on hourly data  
are more scientifically correct.**

# Locations of Southern Taiwan Supersites



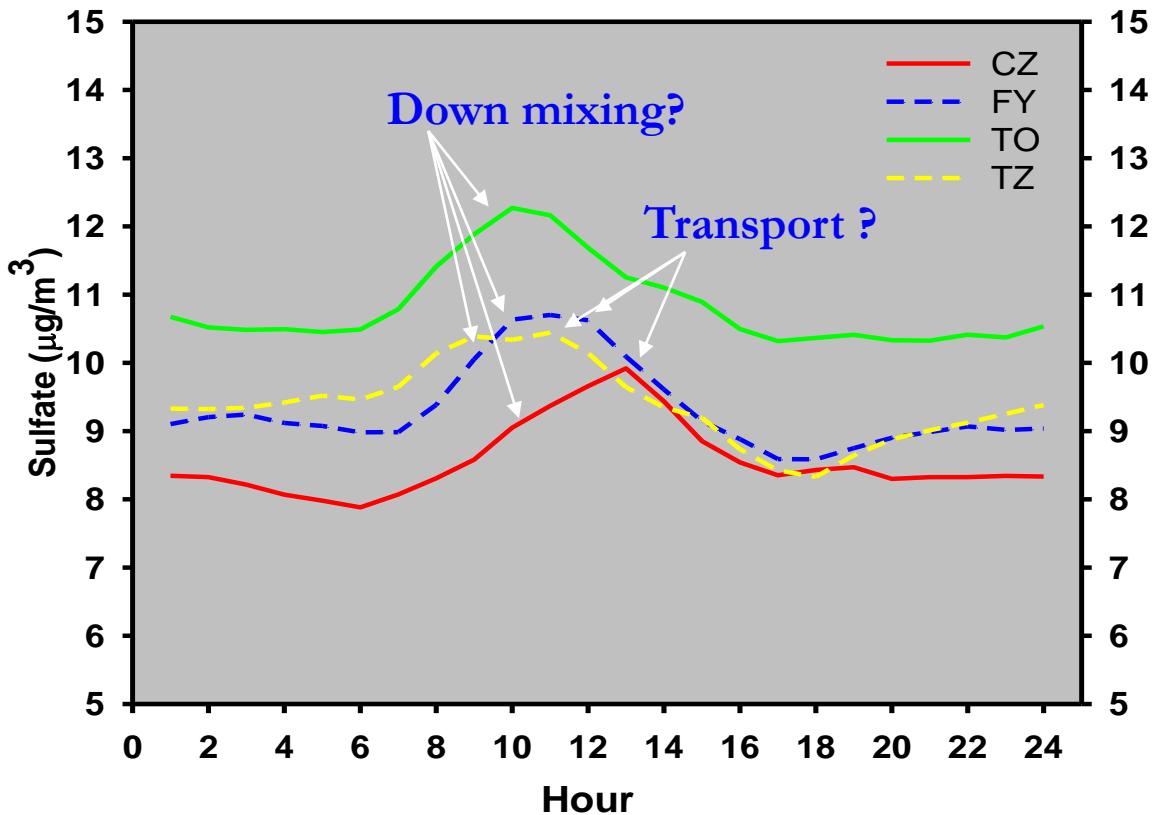
# Purposes of Southern Taiwan PM Supersites Study

- To characterize the ambient PM concentrations, size distributions, and chemical compositions;
- To study the formation pathways of PM;
- To formulate control strategies/measures.

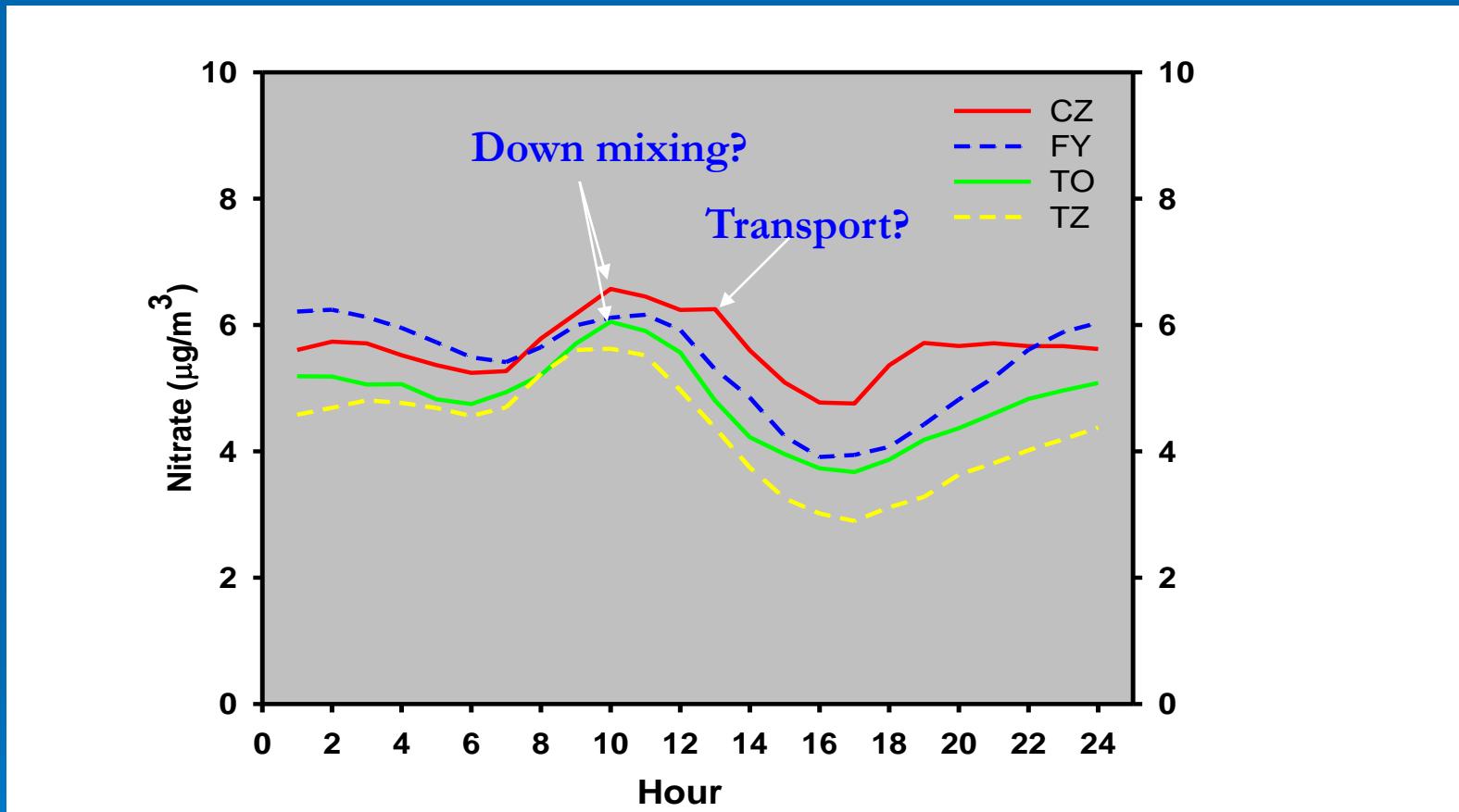
# Measurement Methods

Species	Measurement Method
$\text{SO}_2$	UV fluorescence
$\text{NO}_x$	Chemiluminescence
$\text{O}_3$	UV absorption
PM	RP1400
Nitrate	RP8400N
Sulfate	RP8400S
OC/EC	Sunset 5040
$\text{H}_2\text{O}_2$	AL-2021
$\text{HNO}_3$	RP3500 (Denuder)

# Comparison of diurnal variations at different sites (Sulfate)

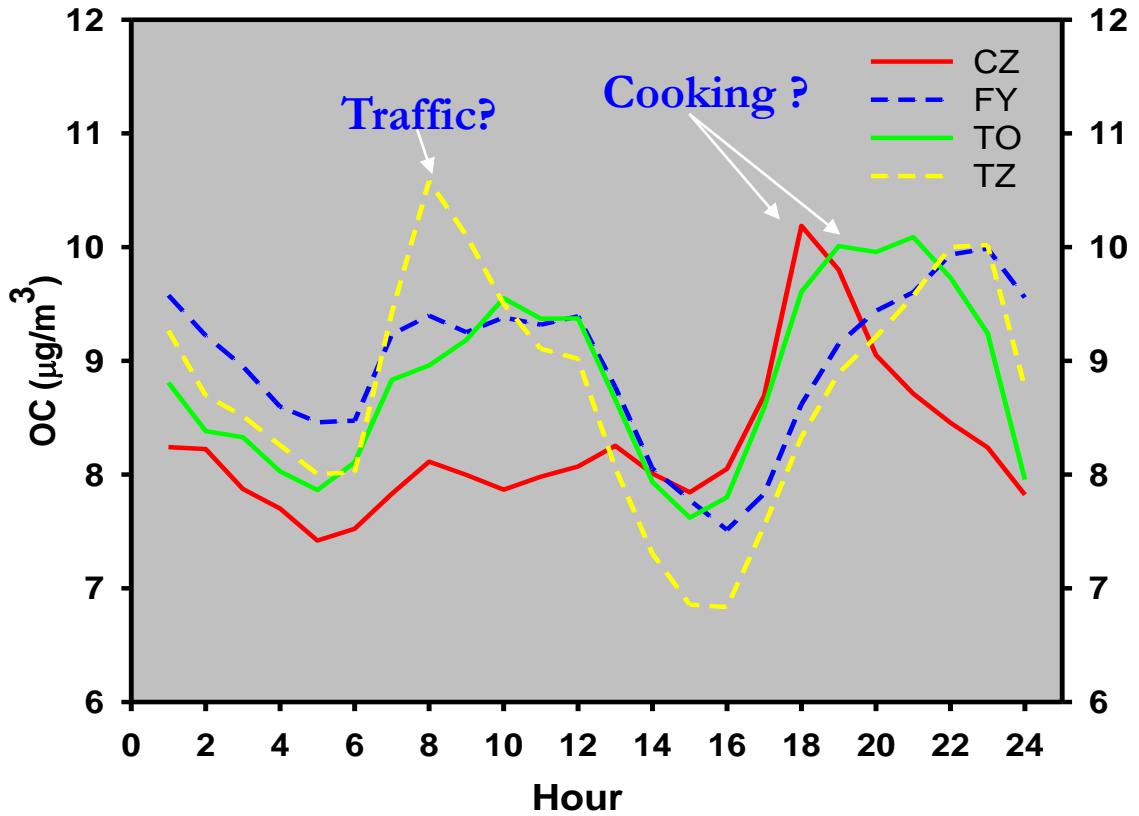


# Comparison of diurnal variations at different sites (Nitrate)

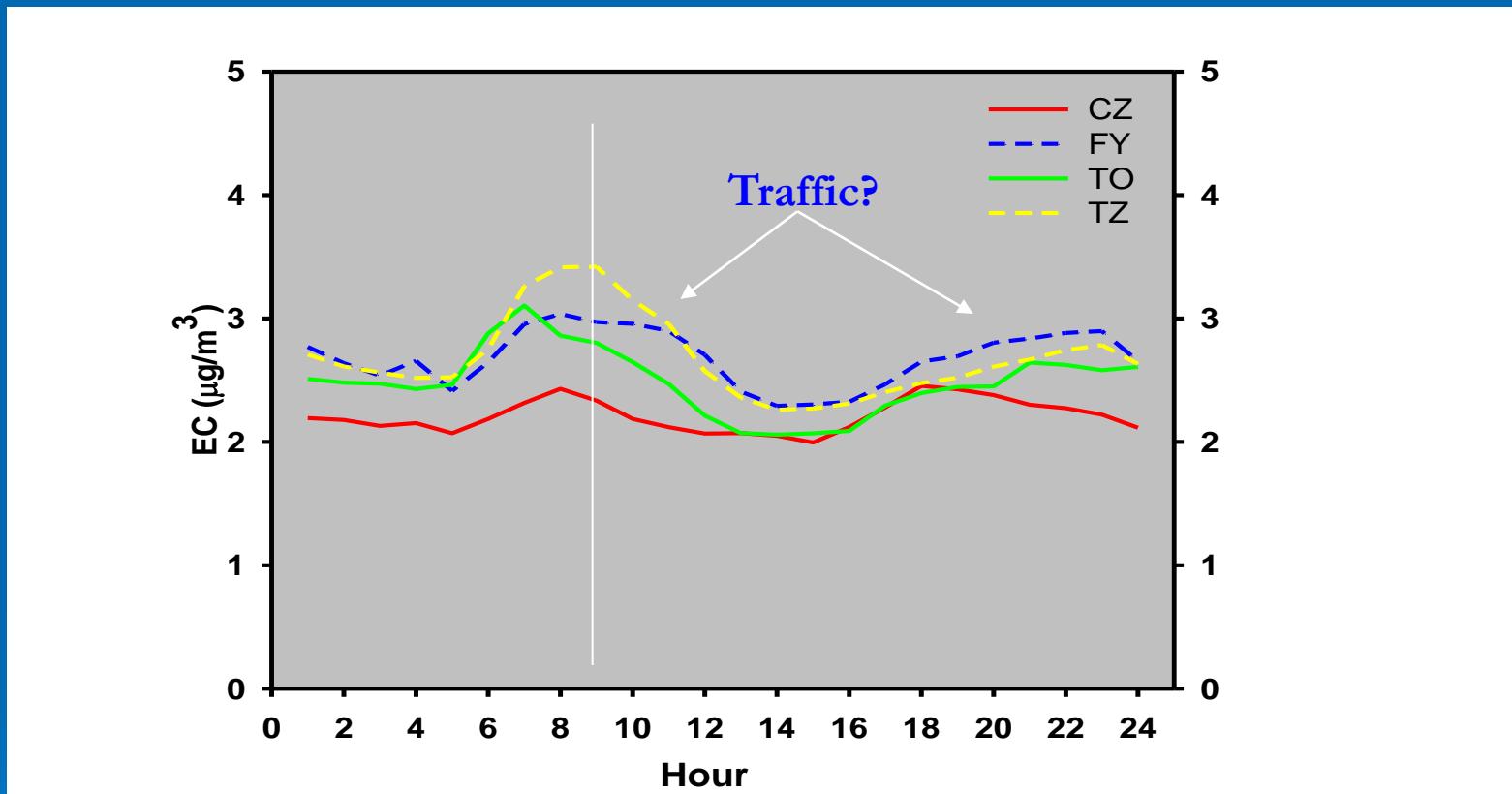


Bottom line: (1) CZ peak different from other stations could be resulted from transport

# Comparison of diurnal variations at different sites (OC)



# Comparison of diurnal variations at different sites (EC)



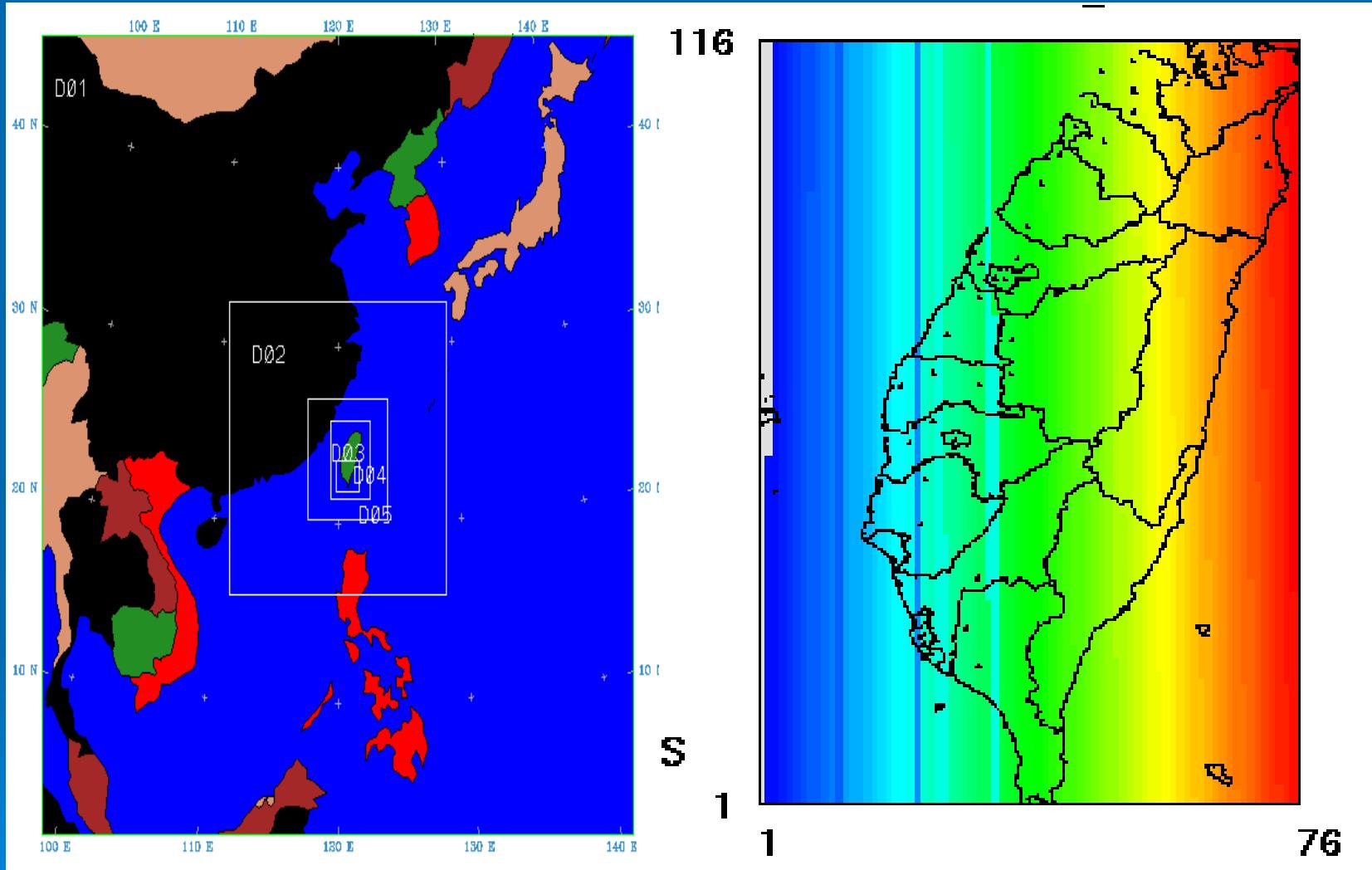
Bottom line: (1) All sites peak at 7-9 h in the morning indicating traffic related.

# CMAQ Modules

- Meteorology : MM5 with MCIP
- Vertical layers: 24 in MM5 and 15 in CCTM
- Chemistry : CB-IV
- Aerosol : Aerosol module version 3
- Simulation periods: 2005/11/01-05

Heteorogeneous HONO formation mechanism added

# Simulation Domains

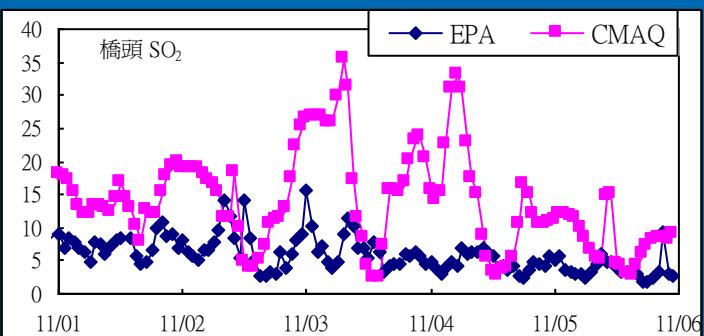


4 layers nesting to 3 km by 3 km

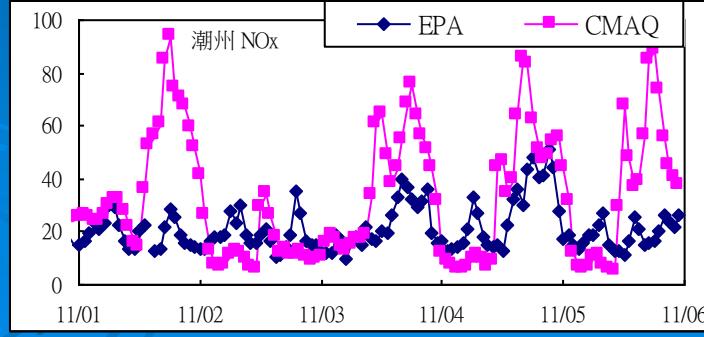
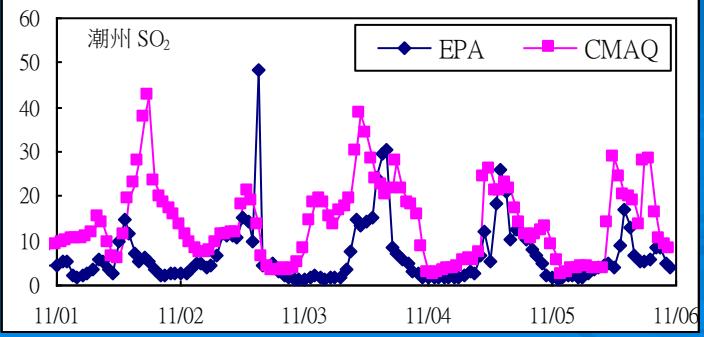
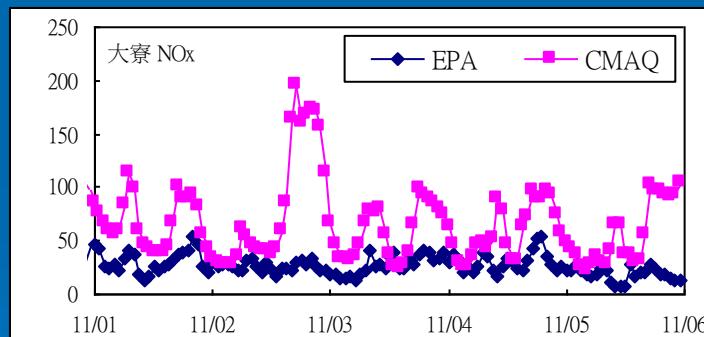
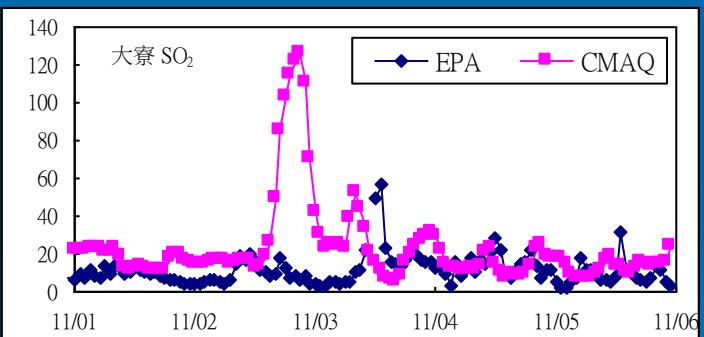
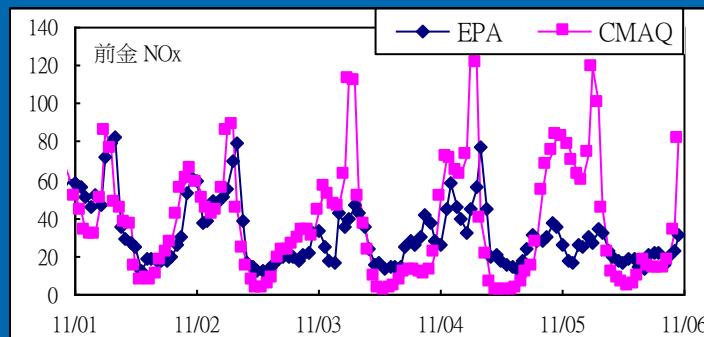
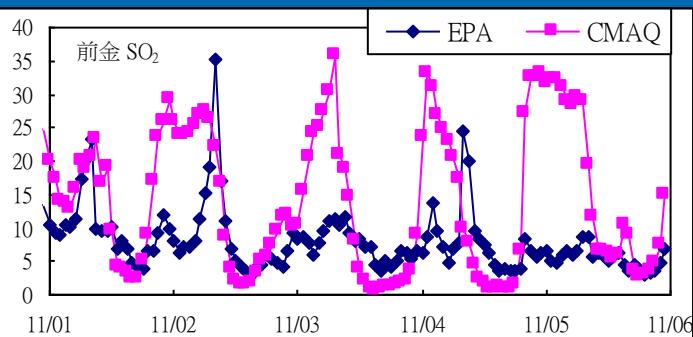
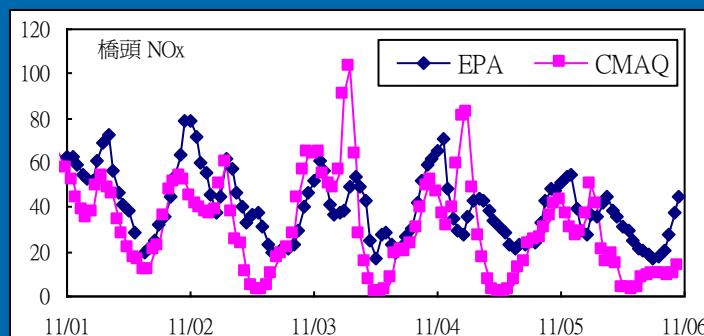
# Sigma-layer

Layer	Sigma	Pressure ( Pa )	Height ( m )	Half Range ( m )
15	0.000	10000	21146	17223
14	0.150	23500	13300	10688
13	0.350	41500	8077	6784
12	0.500	55000	5490	4794
11	0.600	64000	4099	3494
10	0.700	73000	2890	2616
9	<b>0.750</b>	<b>77500</b>	<b>2341</b>	<b>2082</b>
8	<b>0.800</b>	<b>82000</b>	<b>1823</b>	<b>1577</b>
7	<b>0.850</b>	<b>86500</b>	<b>1332</b>	<b>1145</b>
6	<b>0.890</b>	<b>90100</b>	<b>957</b>	<b>777</b>
5	<b>0.930</b>	<b>93700</b>	<b>598</b>	<b>467</b>
4	<b>0.960</b>	<b>96400</b>	<b>337</b>	<b>252</b>
3	<b>0.980</b>	<b>98200</b>	<b>167</b>	<b>125</b>
2	<b>0.990</b>	<b>99100</b>	<b>83</b>	<b>62</b>
1	<b>0.995</b>	<b>99550</b>	<b>41</b>	<b>21</b>
0	1.000	100000	0	

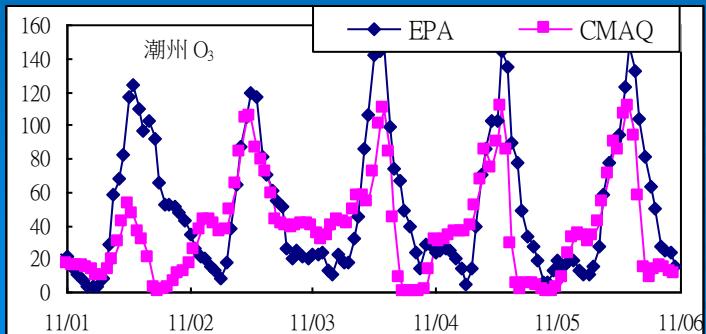
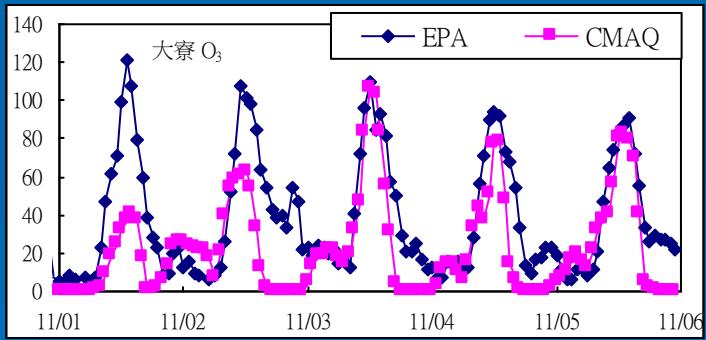
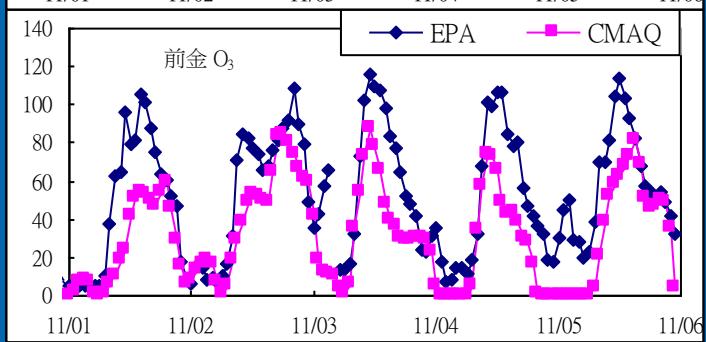
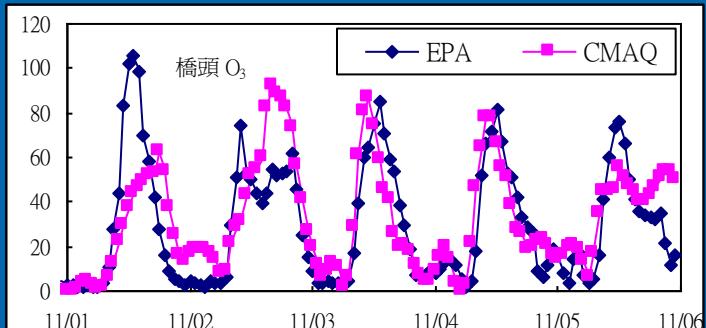
**SO<sub>2</sub>**



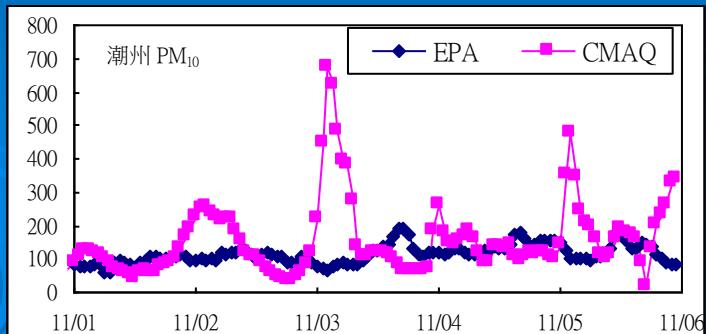
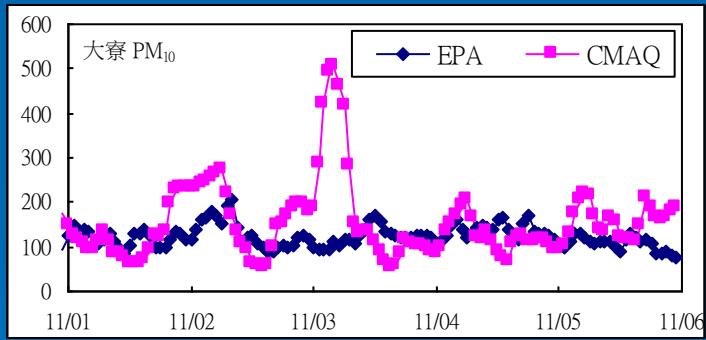
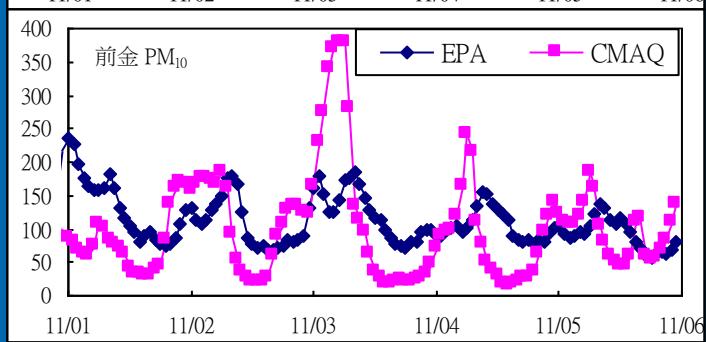
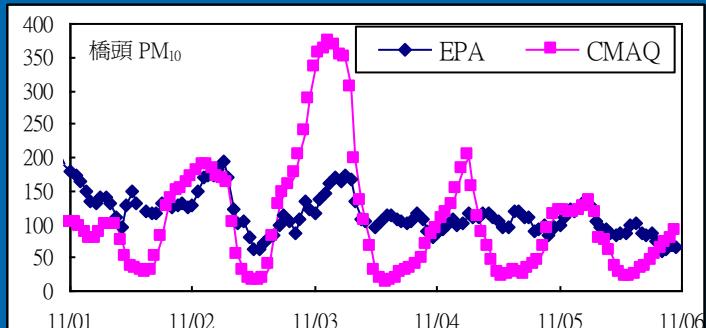
**NOx**



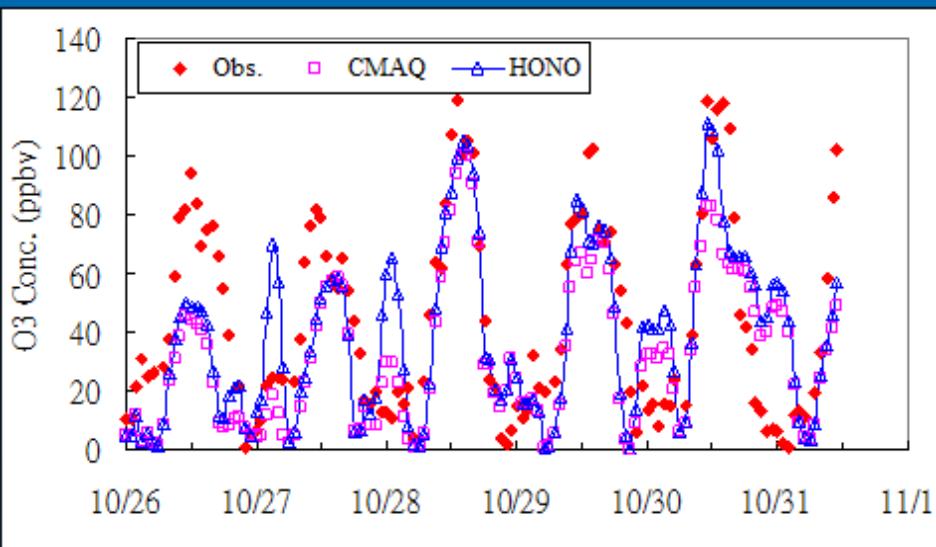
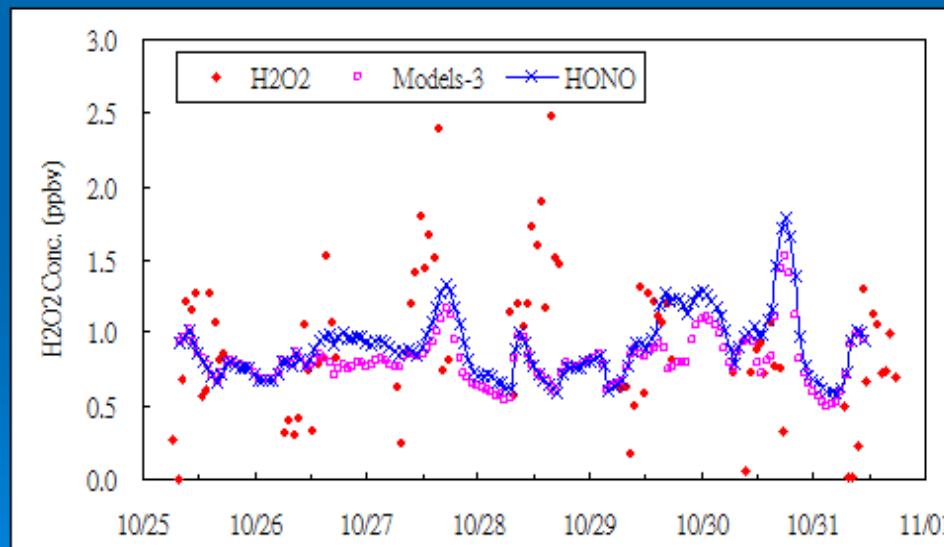
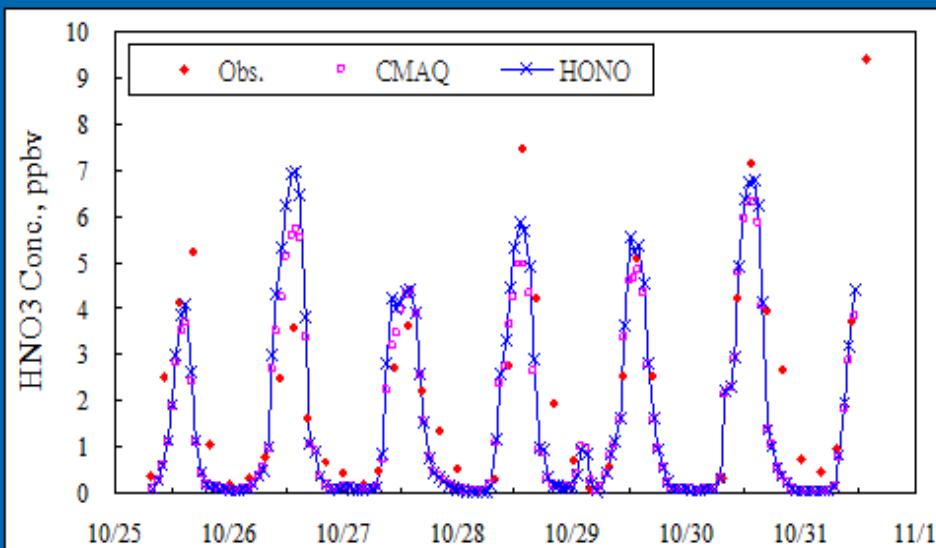
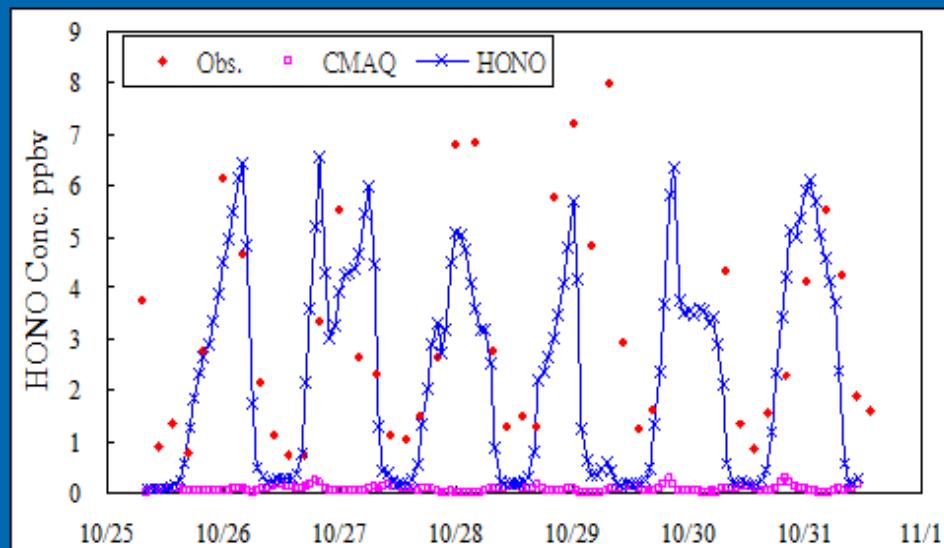
$O_3$



PM

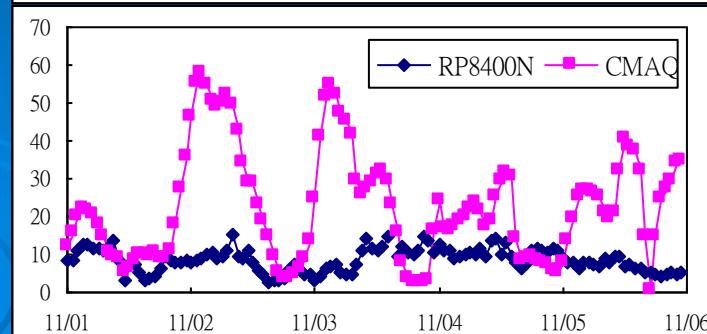
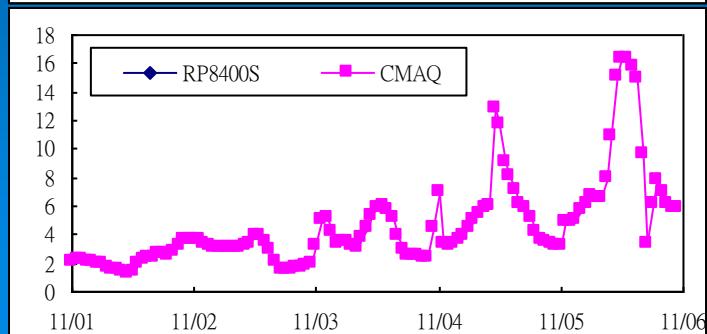
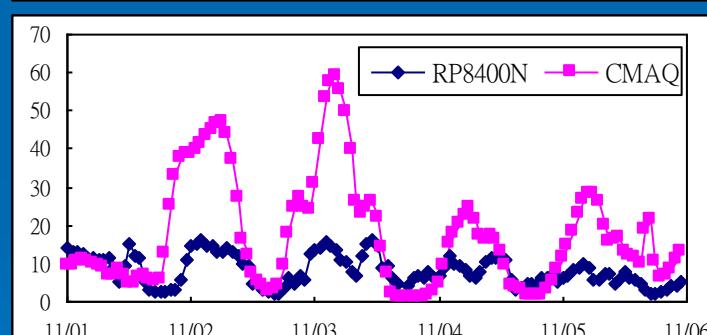
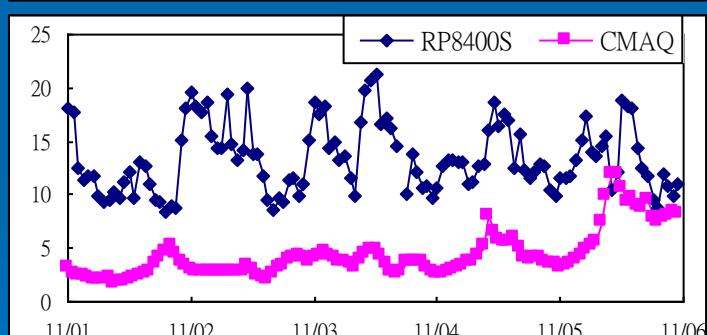
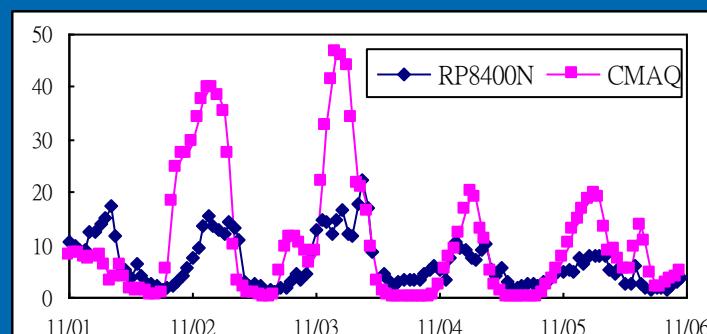
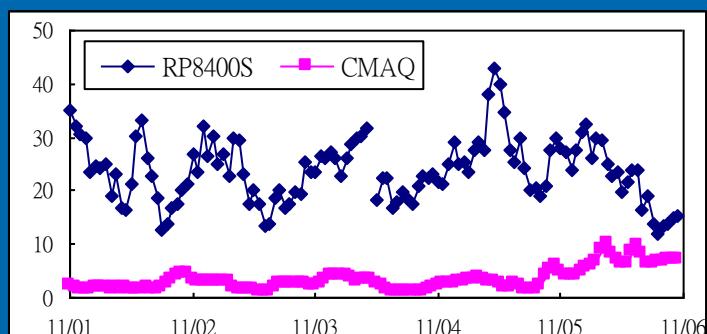
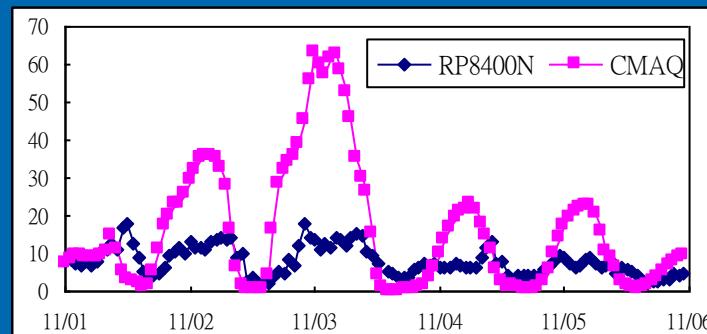
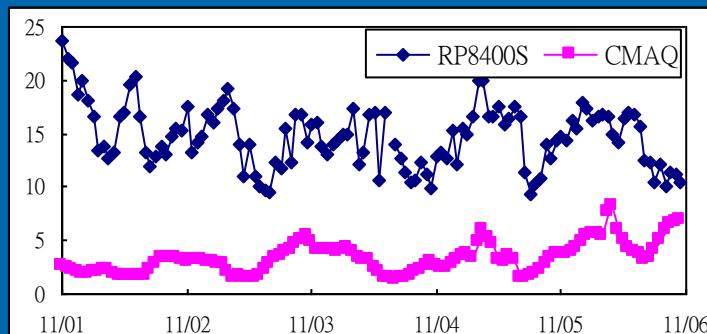


# Modifications of HONO Formation



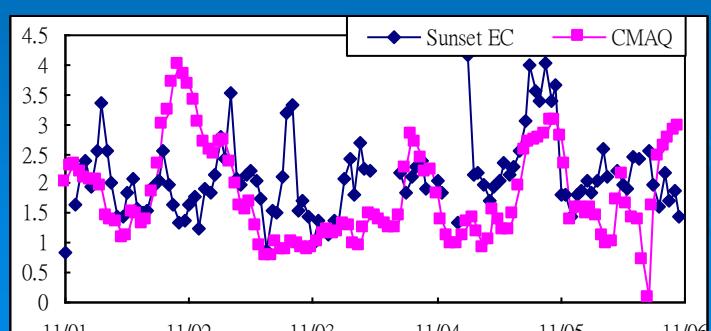
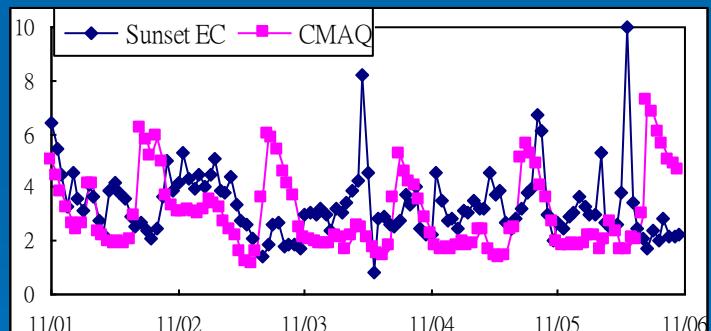
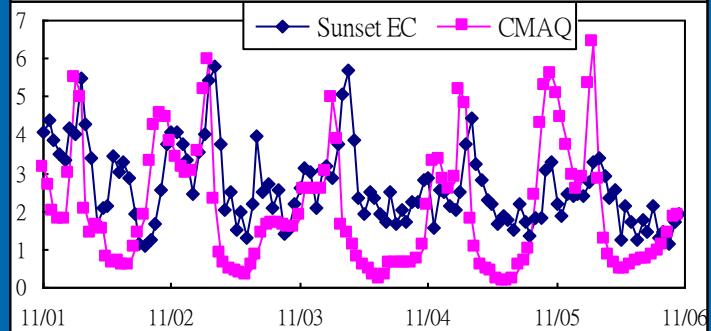
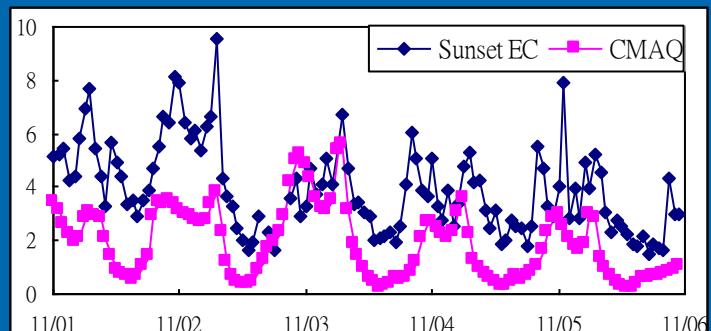
Heterogeneous HONO formation mechanism  
(Ammann et al., 1998; Arens et al., 2001)

## Sulfate

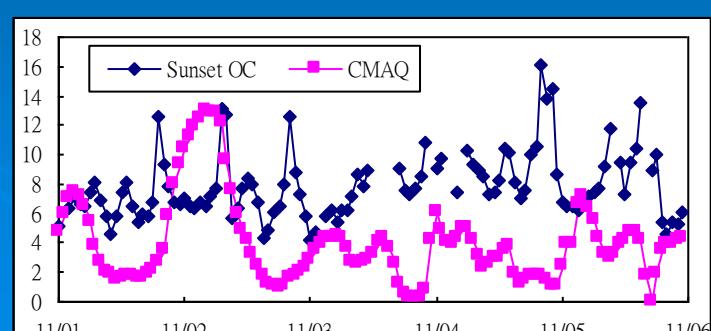
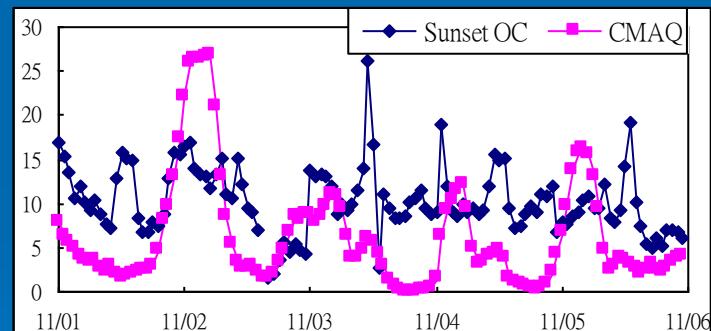
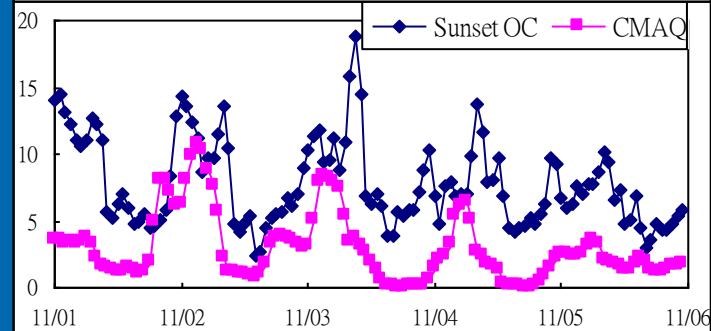
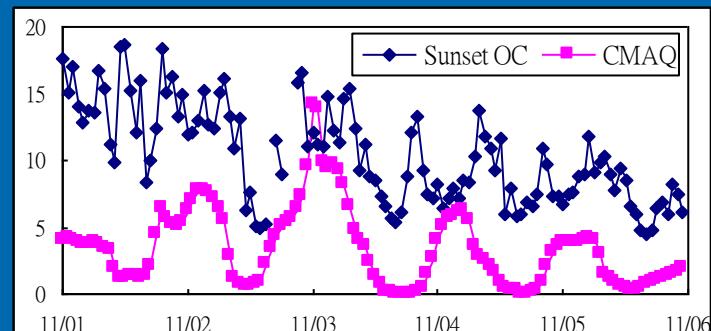


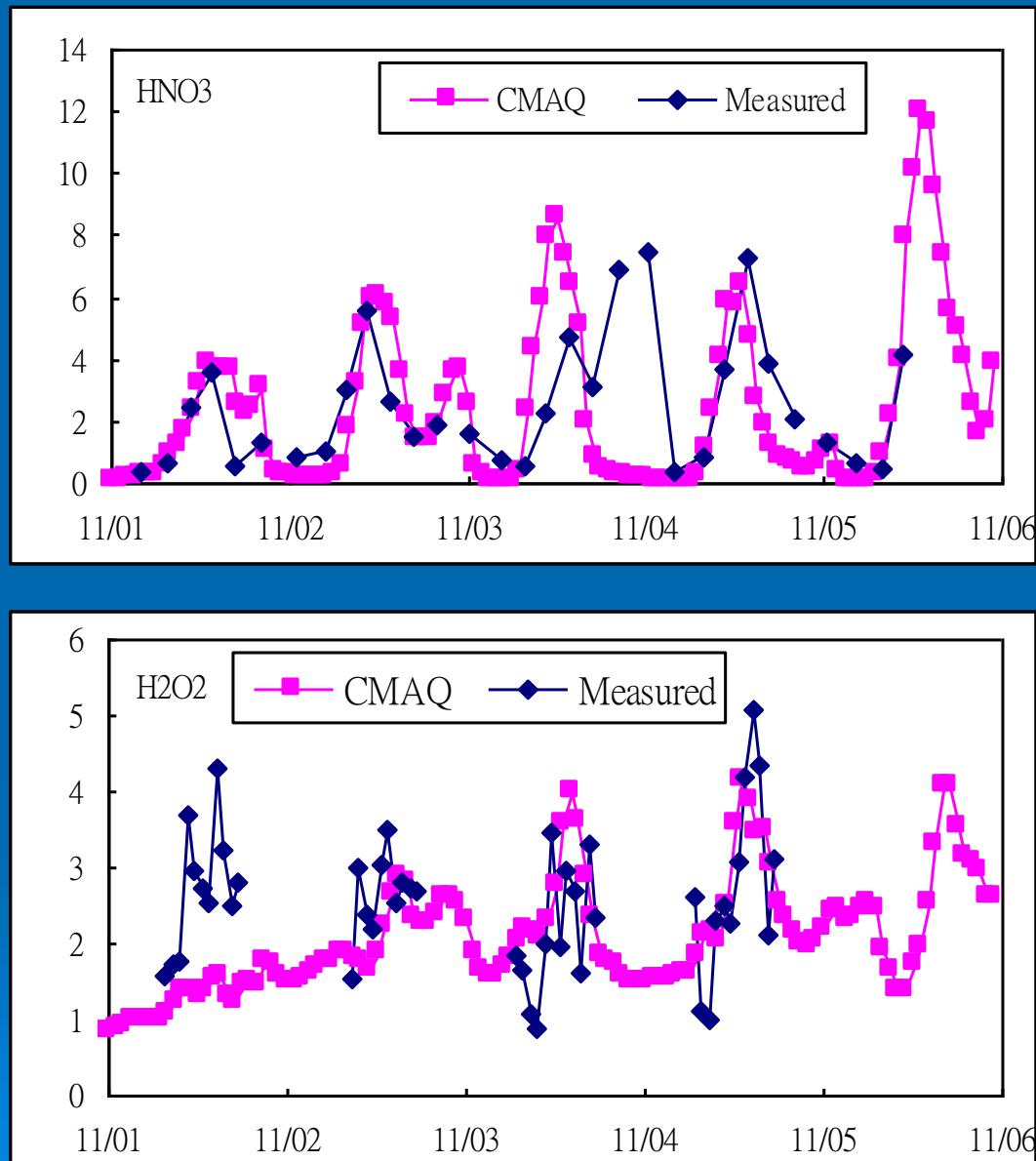
## Nitrate

EC



OC





# Ratios of Simulated to Measured

	North	West	Center	South
SOx	2.6±1.5	1.9±1.5	2.9±3.2	2.9±2.4
NOx	0.76±0.49	1.2±0.89	2.6±1.7	1.6±1.3
O3	0.92±0.38	0.61±0.25	0.53±0.34	0.55±0.35
PM	0.83±0.59	0.86±0.60	1.3±0.81	1.4±1.5
EC	0.46±0.30	0.78±0.61	1.0±0.70	0.95±0.51
OC	0.31±0.26	0.36±0.30	0.63±0.54	0.56±0.44
Nitrate	1.7±1.4	1.5±1.6	2.2±1.9	3.0±2.2
Sulfate	0.23±0.11	0.15±0.11	0.33±0.19	

# Formation of Secondary PM:

## Sulfate PM formation:



Gas Phase:  $O_2, H_2O_2$



Aqueous Phase:  $H_2O$



## Nitrate PM formation:



Gas Phase : (daytime)



Gas & Aq Phase : (nighttime)

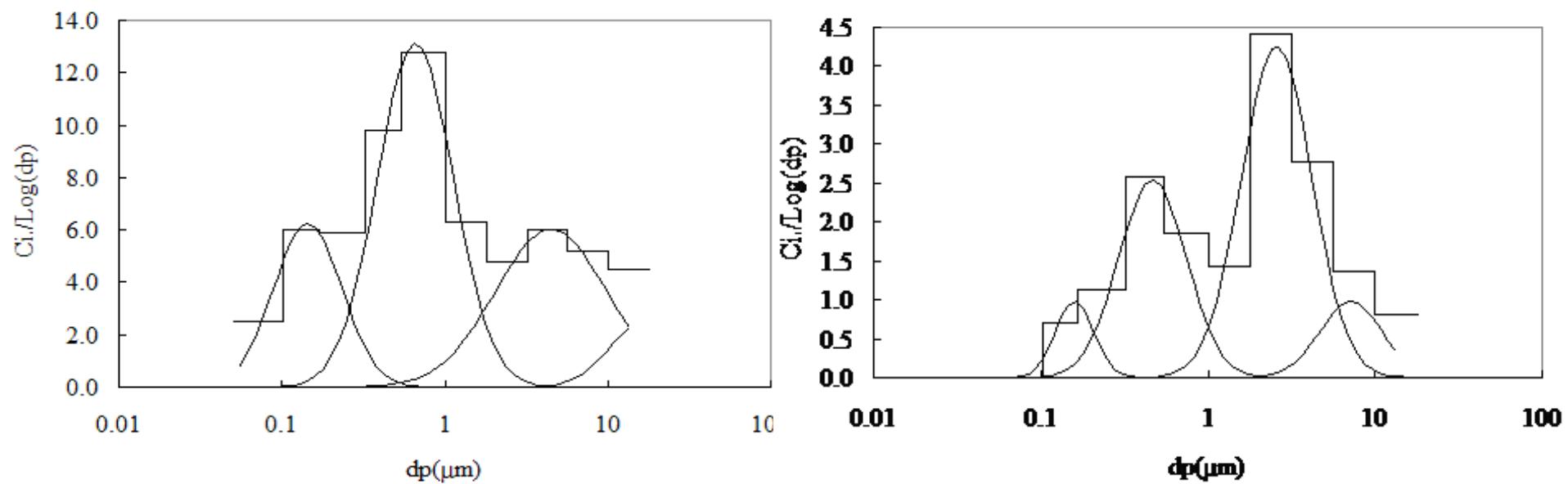


## Organic PM formation: (Gas-Particle conversion)



(Long-chain VOC, Aromatics,  
Biogenic VOC)

# Size Distributions of Nitrate and Sulfate

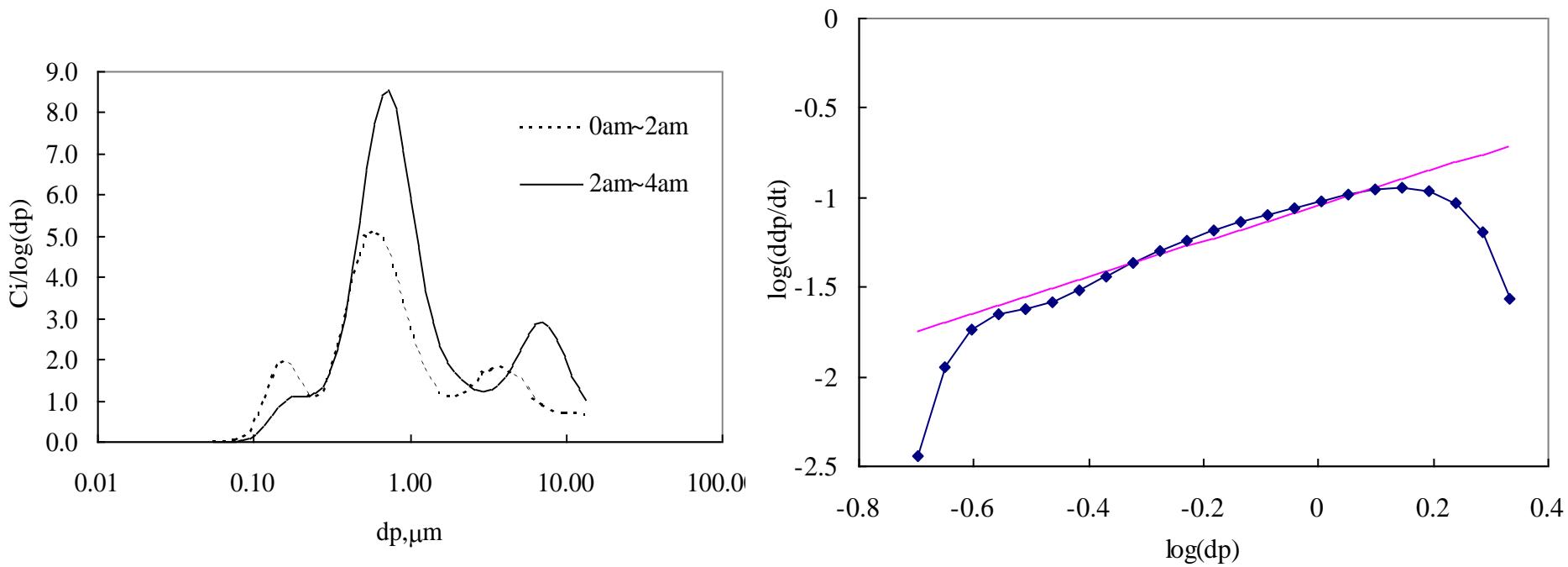


Nitrate

Sulfate

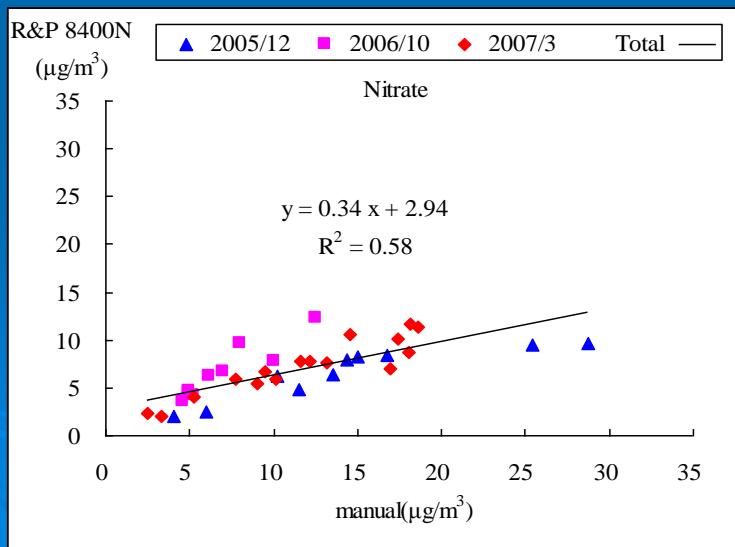
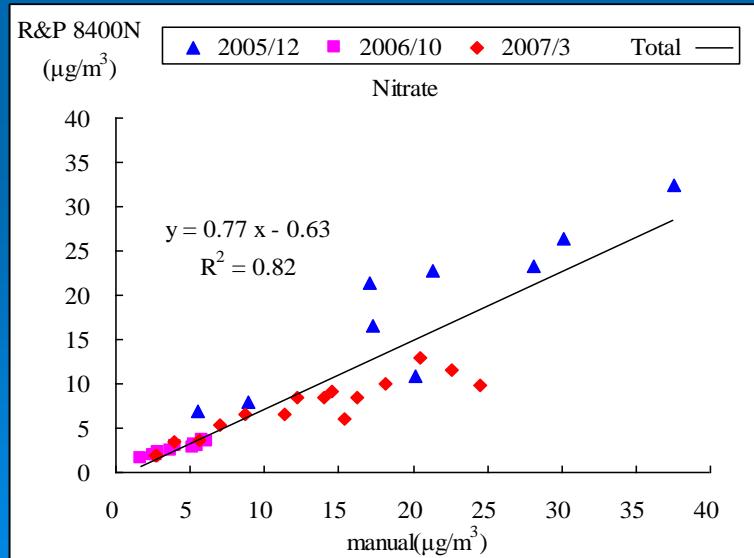
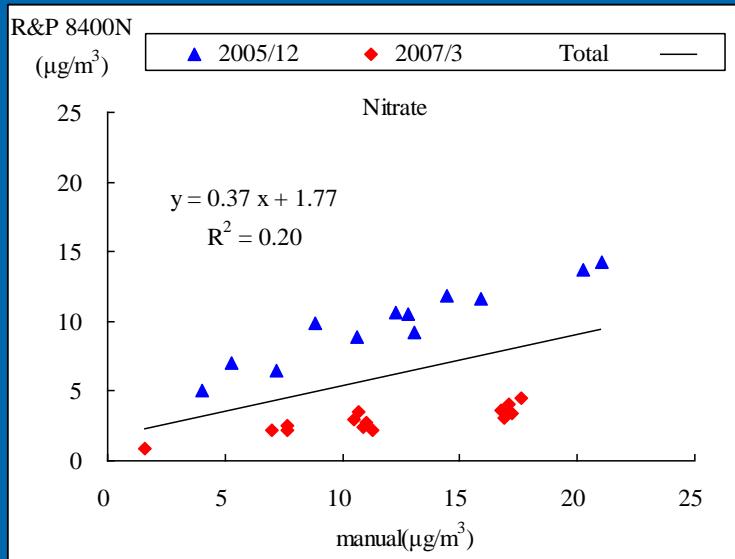
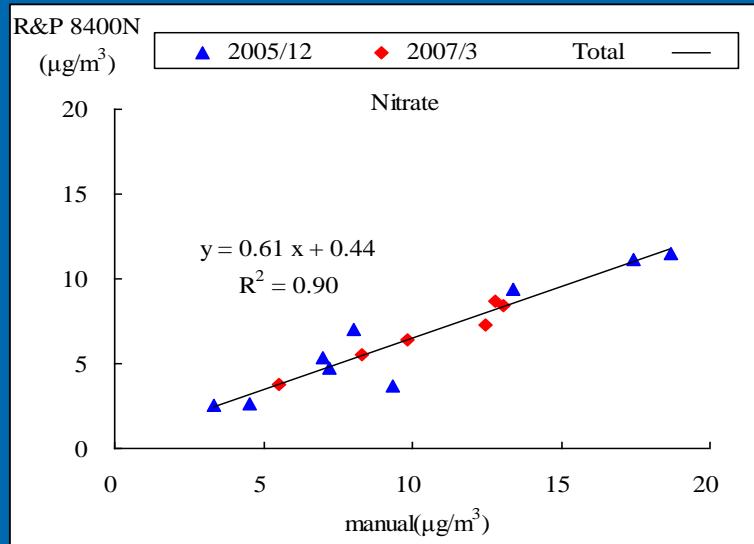
Mode Diameter	Contribution Fraction	Mode Diameter	Contribution Fraction
0.11 to 0.14 (0.12)	13.8 to 45.2 (22.2)	0.11 to 0.12 (0.12)	15.0 to 23.7 (14.3)
0.33 to 0.74 (0.51)	21.8 to 54.8 (41.9)	0.37 to 0.58 (0.47)	30.4 to 58.8 (36.3)
4.27 to 6.00 (4.80)	29.1 to 46.8 (36.2)	1.22 to 1.49 (1.36)	29.3 to 45.9 (28.1)
		3.91 to 4.61 (4.32)	25.8 to 33.5 (21.4)

# Formation Mechanisms of Sulfate

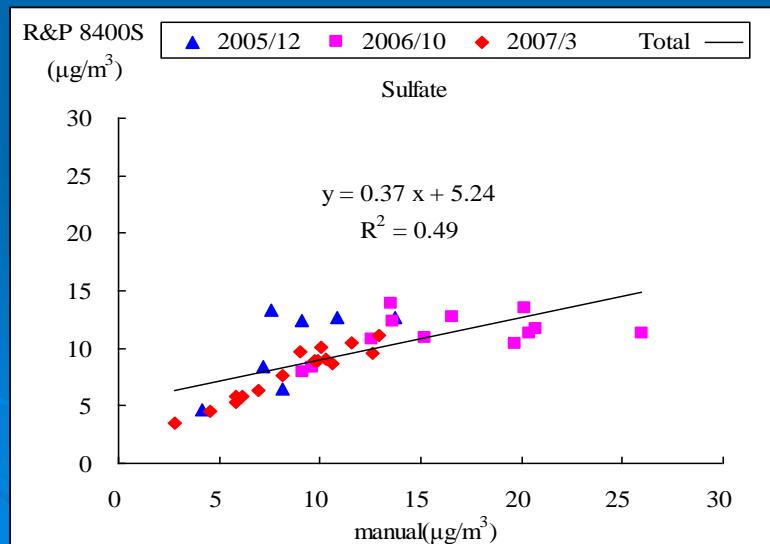
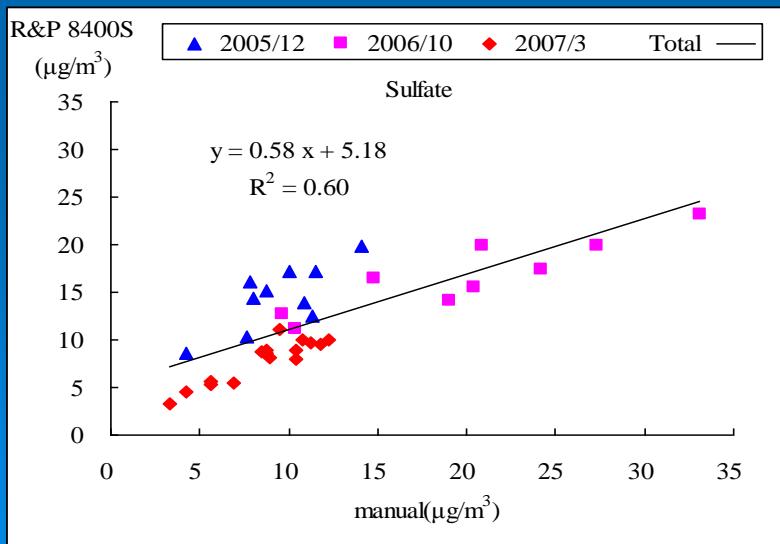
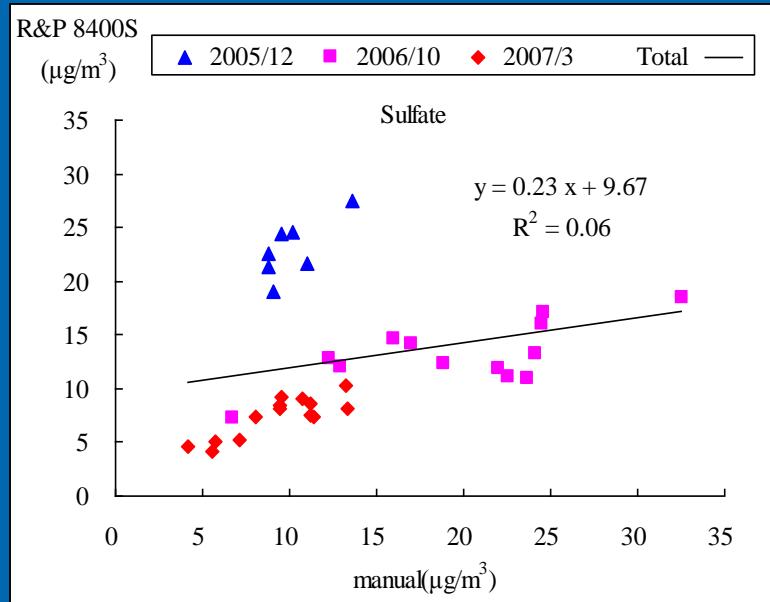
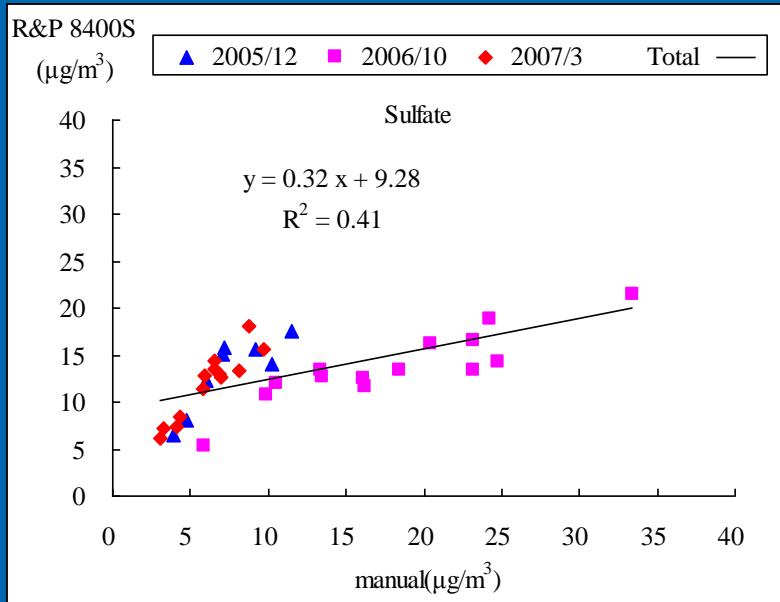


Size distributions for sulfate measured in sequences and the computed particle growth rates. The straight line is for curve with slope of one.

# RP8400N vs. Denuder



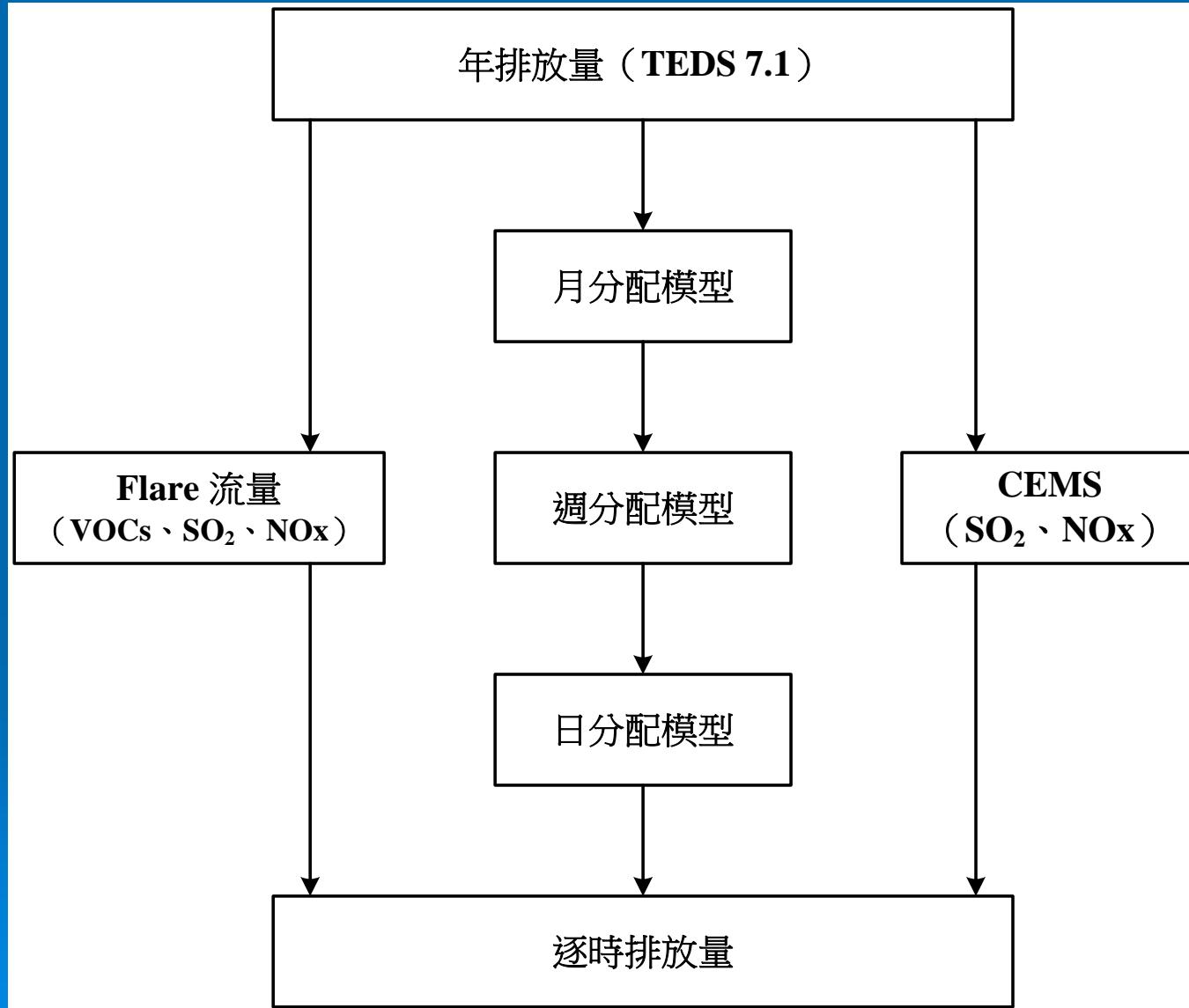
# RP8400S vs. Denuder



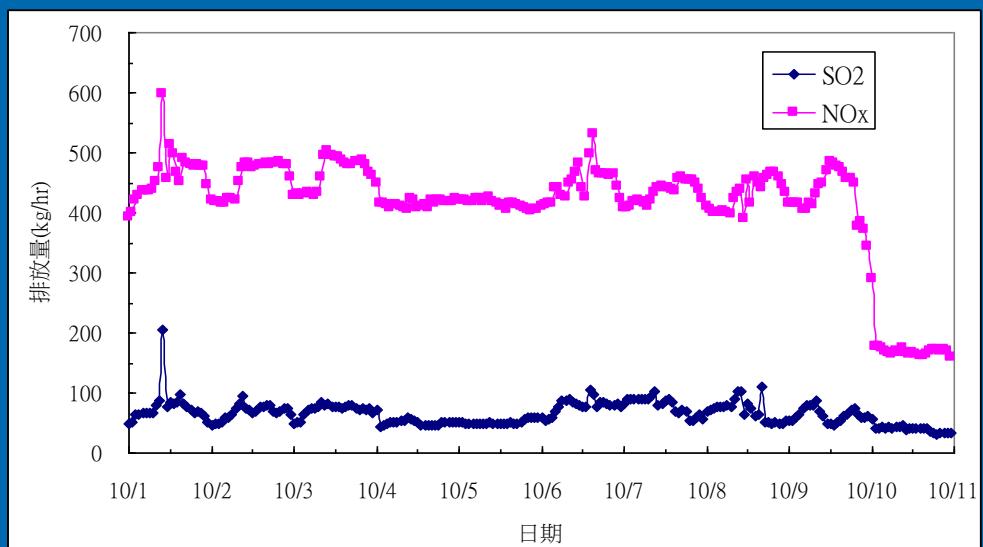
# Theoretical Conversions of RP8400

Site	RP8400S		RP8400N	
	$\text{Na}_2\text{SO}_4$	$(\text{NH}_4)_2\text{SO}_4$	$\text{KNO}_3$	$\text{NH}_4\text{NO}_3$
North	49	23	88	72
West	62	20	84	67
Center	64	31	90	83
South	67	28	84	76

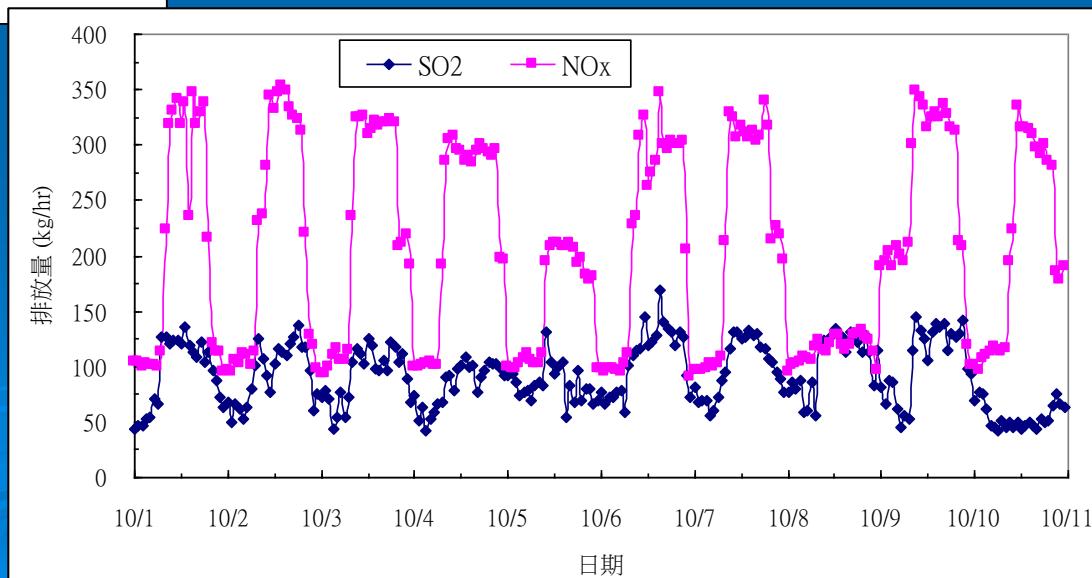
# 建置逐時排放量



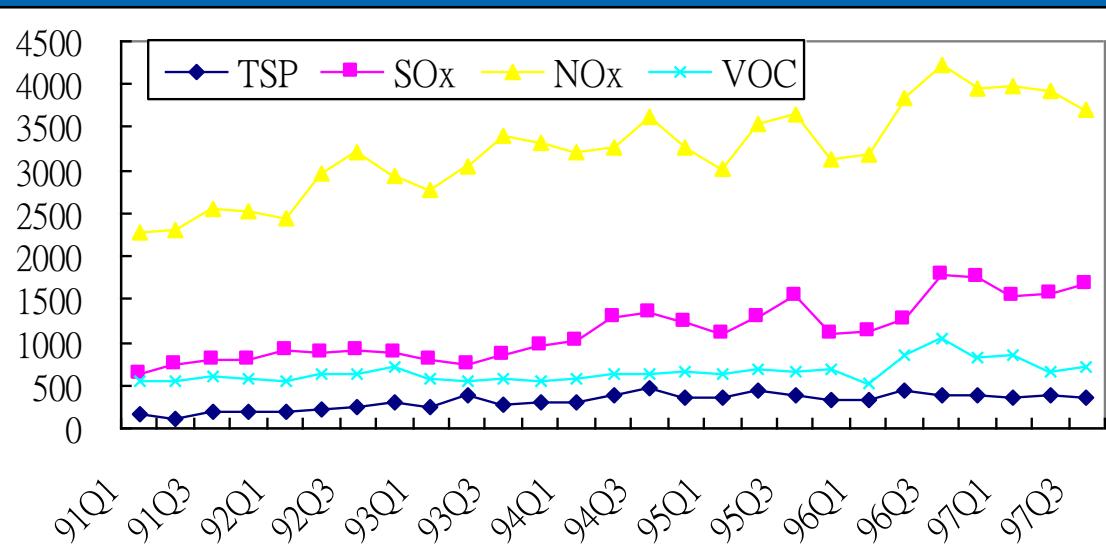
# 嘉義縣與台南縣CEMS之逐時排放量



嘉義縣

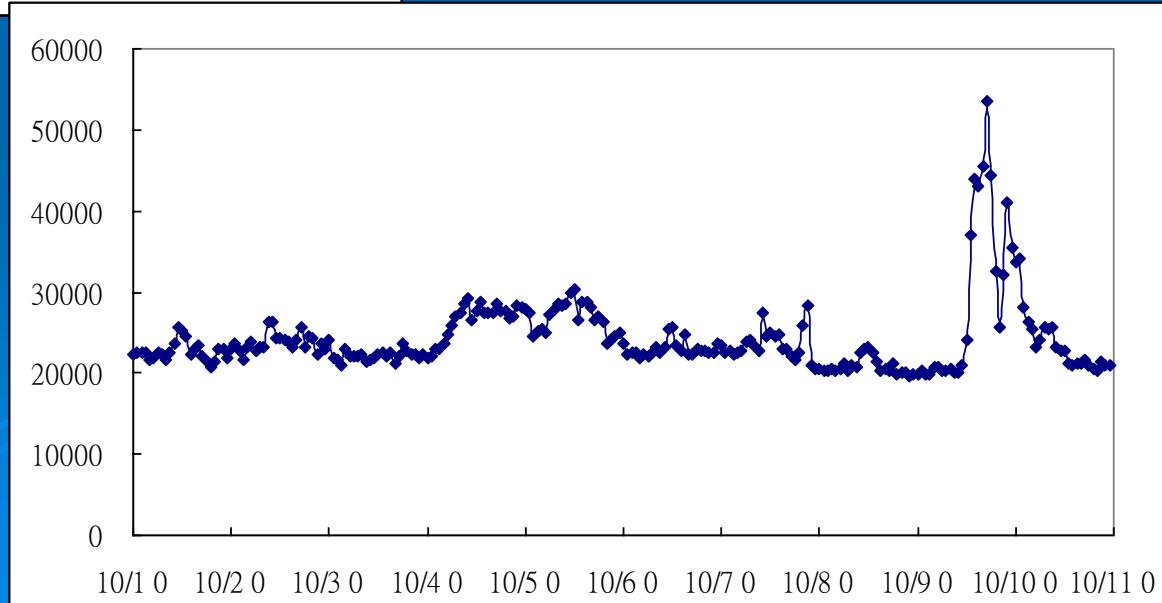


# 離島工業區



廢氣燃燒塔逐時  
流量(Nm<sup>3</sup>/hr)

逐季網路申報  
排放量(公噸)



# Summary

- Ozone concentrations have been simulated by CMAQ fairly;
- The error of simulated PM mass is generally within a factor of two;
- Fair agreements for nitrate and EC;
- Both sulfate and OC are underestimated.

*Conversion of S(IV) to S(VI)  
in aerosol water*

# Suggestions

- Formation of sulfate in aerosol water should be included;
- Better measurement methods for hourly species concentrations should be developed;
- Temporal variations of hourly emission profiles should be developed.

*Thanks for Your  
Attentions.*