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Facile Synthesis of Neodymium Stannate Nanoparticles an Effective Electrocatalyst for the Selective Detection of Dimetridazole in Biological Samples

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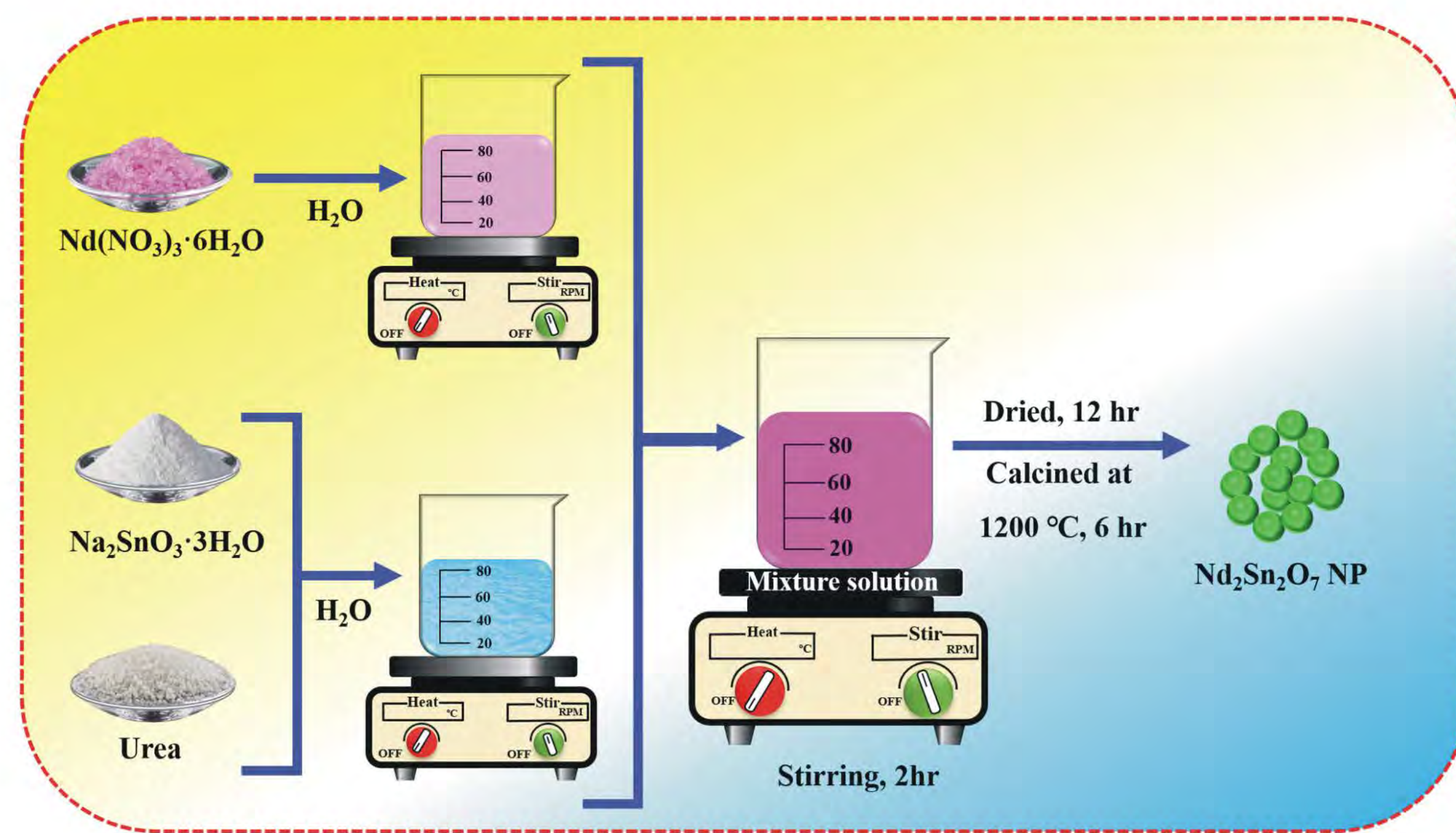
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Abstract

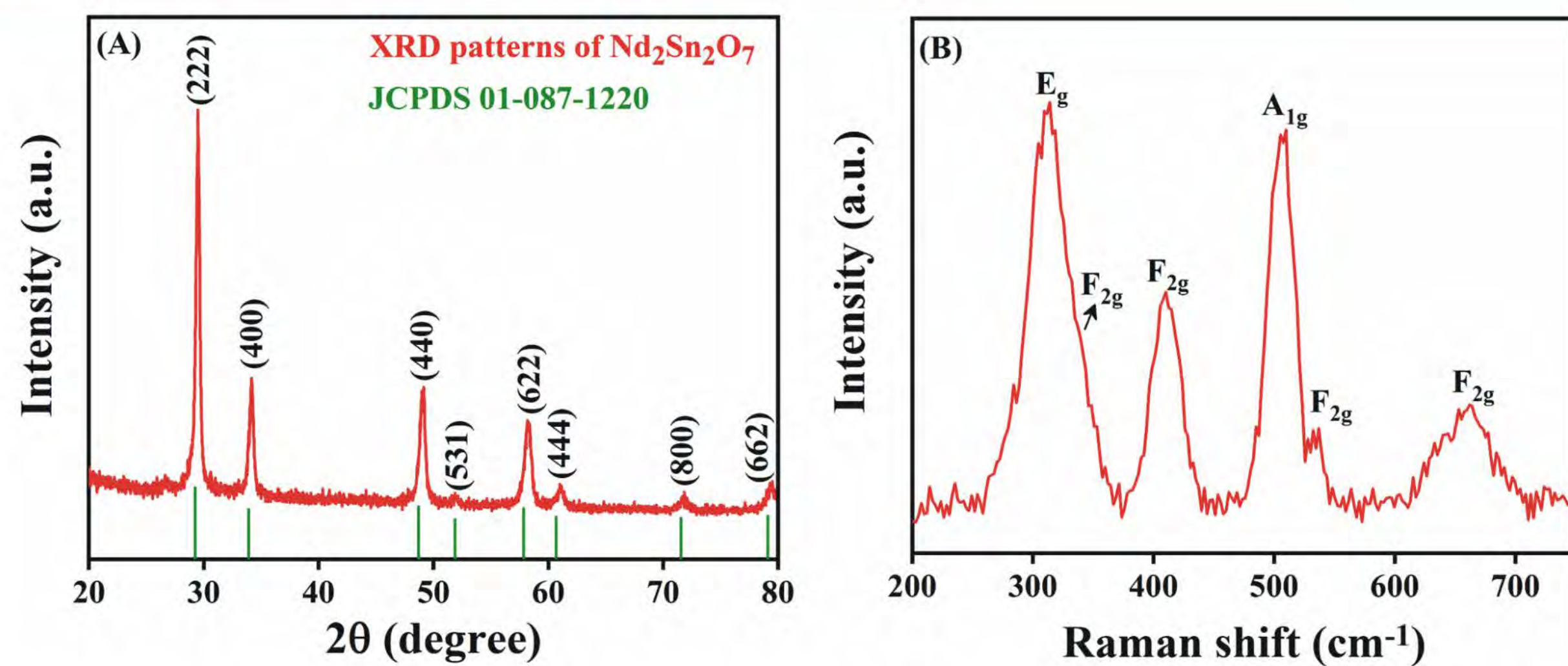
- In this work, pyrochlore neodymium stannate nanoparticles ($\text{Nd}_2\text{Sn}_2\text{O}_7$ NP) have been synthesized by a facile co-precipitation technique and employed as an electrode material on GCE for the determination of dimetridazole (DM) drug.
- The fabricated $\text{Nd}_2\text{Sn}_2\text{O}_7$ NP/GCE shows a lower LOD of 6 nm towards the determination of DM and the calculated sensitivity is $0.61 \mu\text{A} \mu\text{M}^{-1} \text{cm}^{-2}$.
- In addition to that, the constructed sensor delivers notable repeatability, reproducibility, and superior selectivity.
- Furthermore, $\text{Nd}_2\text{Sn}_2\text{O}_7$ NP/GCE sensor displays acceptable recovery results in the real sample analysis in biological samples.

Synthesis Procedure



Scheme 1. The coprecipitation synthesis of $\text{Nd}_2\text{Sn}_2\text{O}_7$ NP.

Results and Discussion



Scheme 1. Fig. 1. (A, B) XRD patterns, and Raman spectra of $\text{Nd}_2\text{Sn}_2\text{O}_7$.

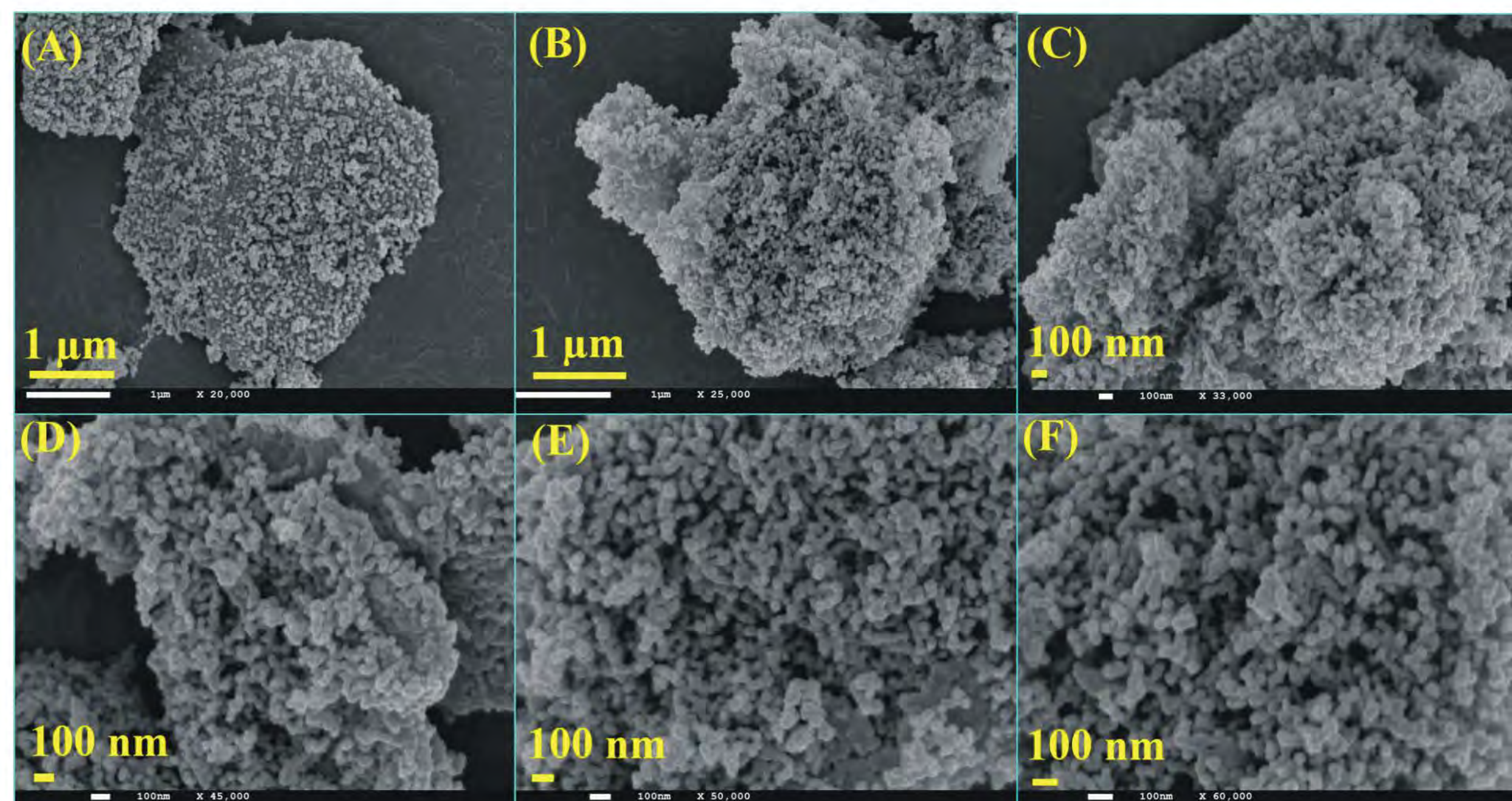


Fig. 2. (A-B) low and (C-F) high magnification FE-SEM images of $\text{Nd}_2\text{Sn}_2\text{O}_7$ NP.

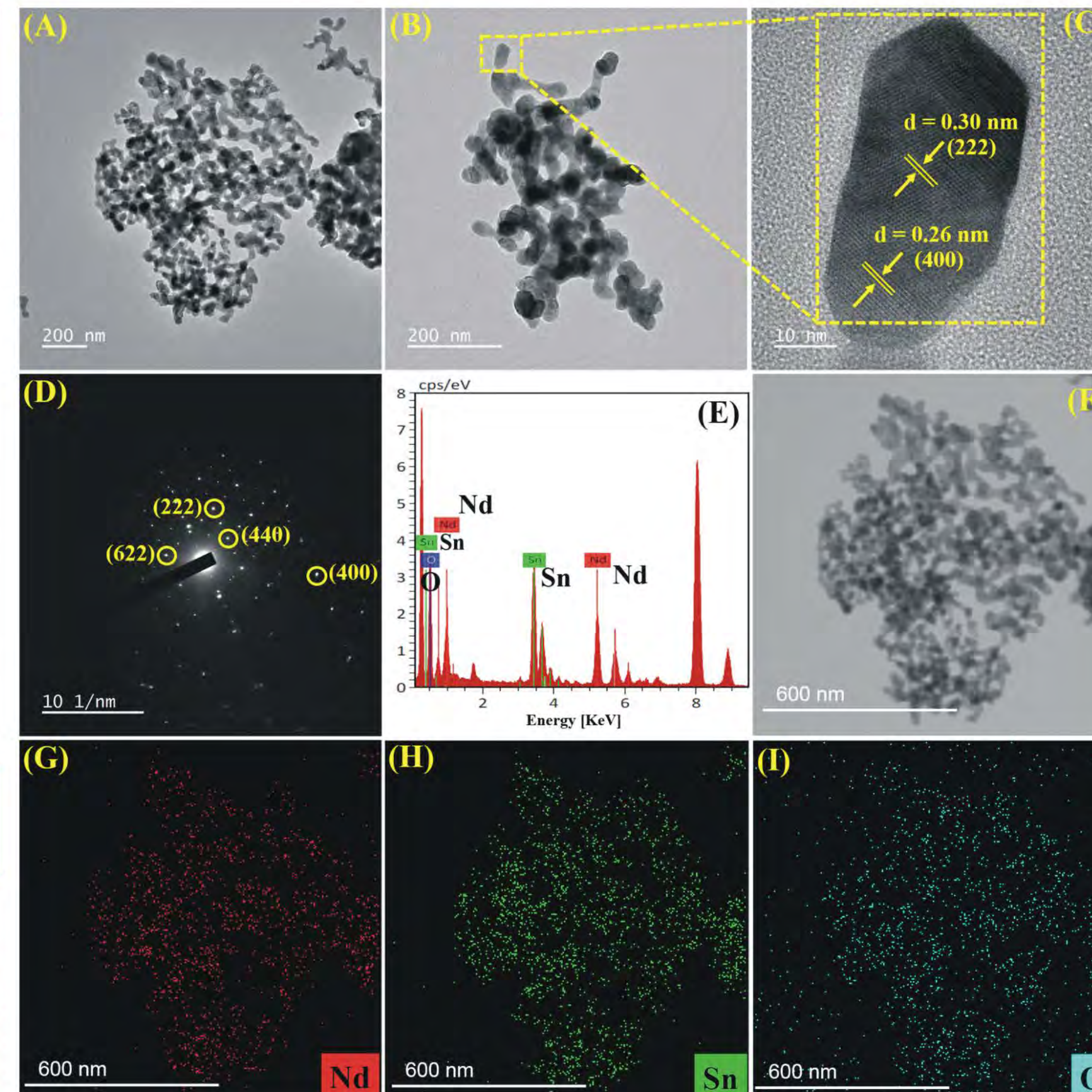


Fig. 3. (A, B) The low-magnification HR-TEM images of $\text{Nd}_2\text{Sn}_2\text{O}_7$. (C) The lattice fringes, (D) SAED patterns, (E) EDX spectrum of $\text{Nd}_2\text{Sn}_2\text{O}_7$ NP. (F) The HR-TEM electron image and (G-I) the corresponding elemental mapping images showing the distribution of elements, namely (G) Neodymium (Nd), (H) Tin (Sn) and (I) Oxygen (O).

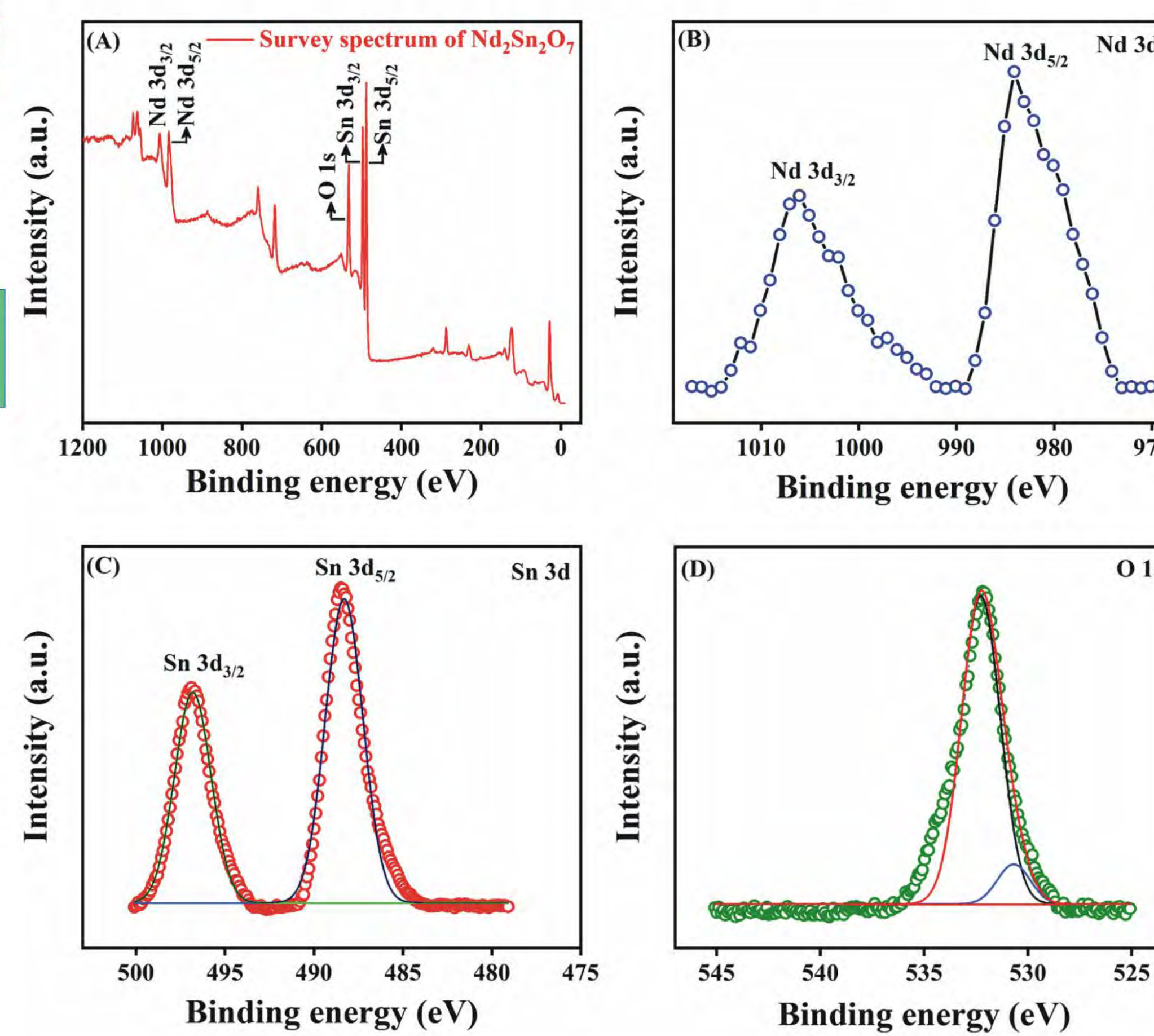
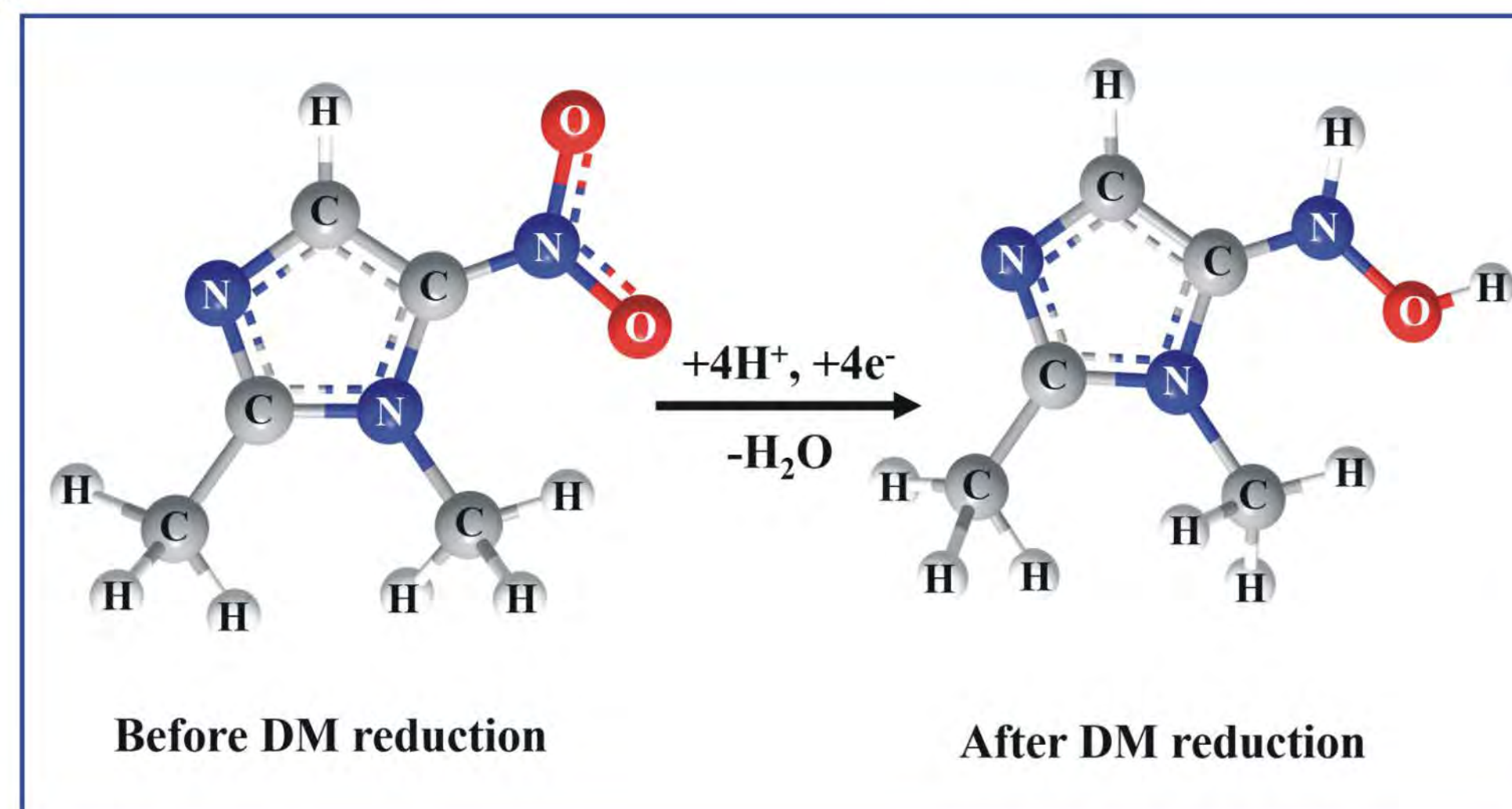


Fig. 4. (A) The overall XPS survey spectrum of $\text{Nd}_2\text{Sn}_2\text{O}_7$ NP. The core-level XPS spectra of Nd 3d (B), Sn 3d (C), and O 1s (D).



Scheme 2. The electrocatalytic reduction mechanism of DM at $\text{Nd}_2\text{Sn}_2\text{O}_7$ modified GCE.

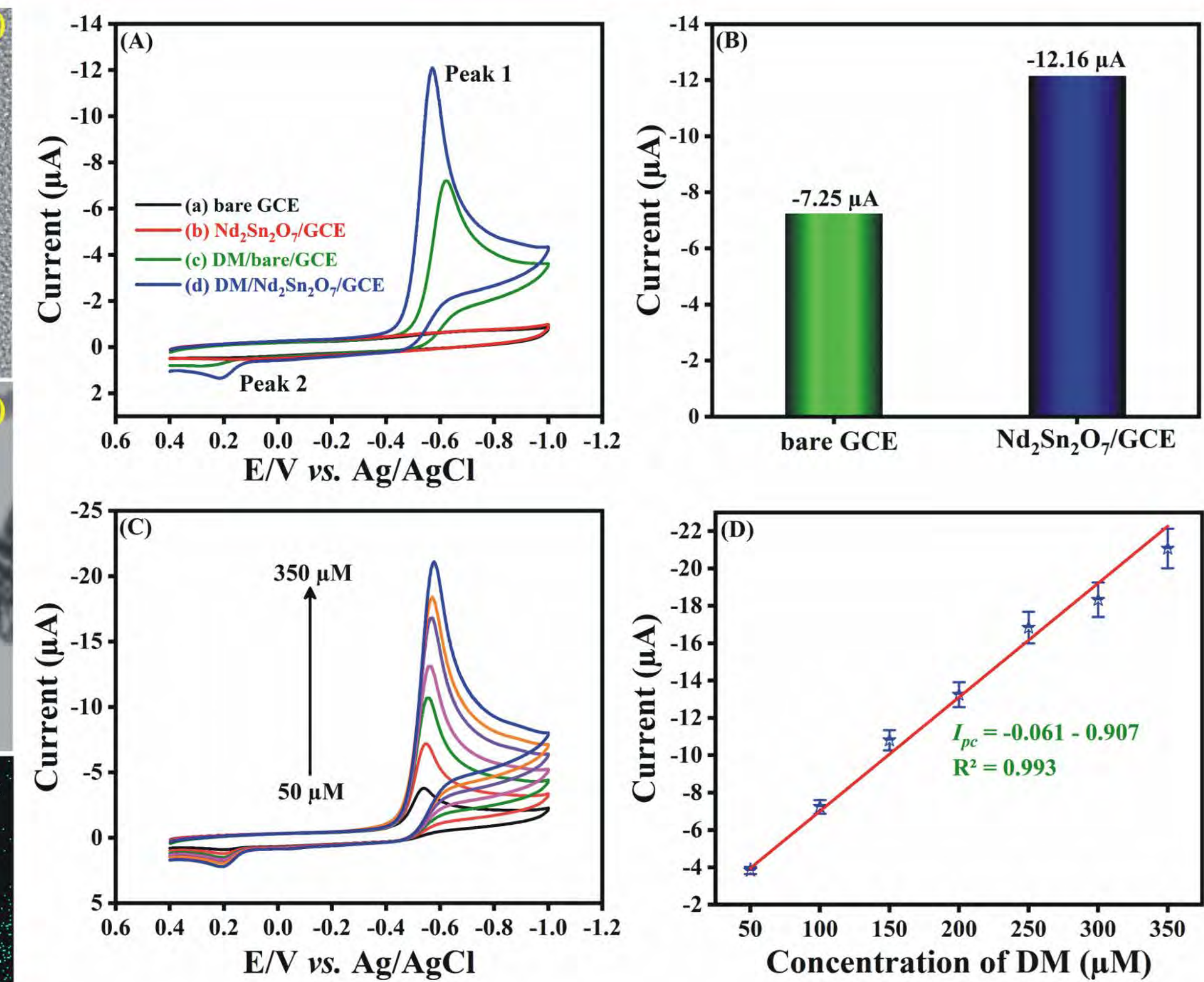


Fig. 5. (A) CV curves of (a) blank GCE, (b) $\text{Nd}_2\text{Sn}_2\text{O}_7$ /GCE in the absence of DM; (c) DM/bare/GCE, (d) DM/ $\text{Nd}_2\text{Sn}_2\text{O}_7$ /GCE in the presence of $150 \mu\text{M}$ of DM. (B) Bar diagram; Recorded cathodic current response of $150 \mu\text{M}$ on bare GCE and $\text{Nd}_2\text{Sn}_2\text{O}_7$ /GCE. (C) The obtained CV signals for different concentrations of DM (50 to $350 \mu\text{M}$) on $\text{Nd}_2\text{Sn}_2\text{O}_7$ /GCE. (D) The linear plot: Concentration of DM (μM) vs. cathodic current (I_{pc}). All experiments were performed at 50 mV/s (Electrolyte: 0.05 M PB solution (pH 7.0)).

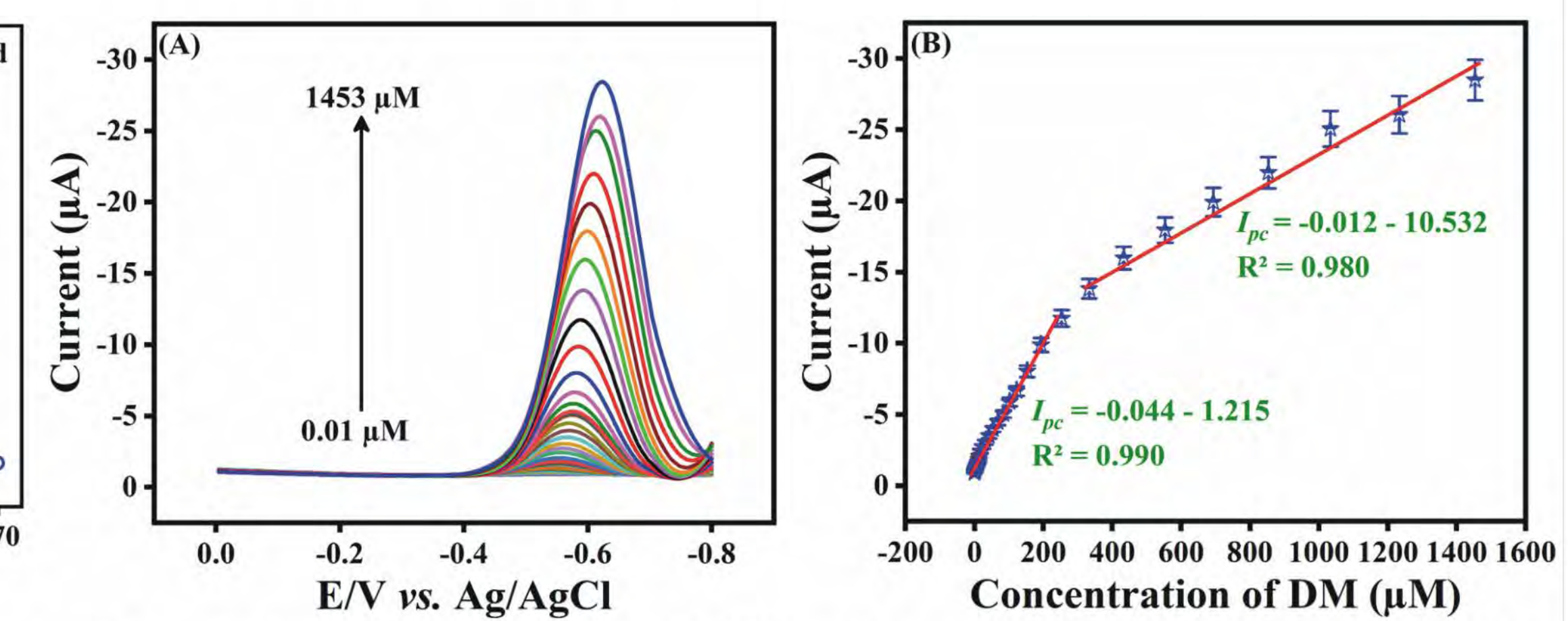


Fig. 6. (A) DPV curves of different concentrations of DM (0.01 to $1453 \mu\text{M}$) in 0.05 M PB solution. (B) The linear calibration plot between the concentrations of DM vs. cathodic current response.

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Conclusion

- In conclusion, we have successfully synthesized a cubic pyrochlore $\text{Nd}_2\text{Sn}_2\text{O}_7$ NP using a low-cost co-precipitation synthesis method.
- The constructed $\text{Nd}_2\text{Sn}_2\text{O}_7$ /GCE shows magnificent electrocatalytic performance towards the determination of veterinary and antibacterial drug DM with a lower LOD of 6 nm .
- From the observed results and electrochemical findings, we believe that the as-prepared highly crystalline cubic pyrochlore $\text{Nd}_2\text{Sn}_2\text{O}_7$ NP/GCE will be implemented to a promising electrochemical sensor for the determination of DM along with the real-sample analyses in biological fluids.